## RESEARCH ARTICLE



# Influence of nitrogen form on the phytoextraction of cadmium by a newly discovered hyperaccumulator *Carpobrotus rossii*

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Received: 6 June 2015 / Accepted: 11 August 2015 / Published online: 10 September 2015 © Springer-Verlag Berlin Heidelberg 2015

**Abstract** Using hyperaccumulator plants is an important method to remove heavy metals from contaminated land. Carpobrotus rossii, a newly found Cd hyperaccumulator, has shown potential to remediate Cd-contaminated soils. This study examined the effect of nitrogen forms on Cd phytoextraction by C. rossii. The plants were grown for 78 days in an acid soil spiked with 20 mg Cd kg<sup>-1</sup> and supplied with (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub>, Ca(NO<sub>3</sub>)<sub>2</sub>, urea, and chicken manure as nitrogen (N) fertilizers. Nitrification inhibitor dicyandiamide (DCD) was applied to maintain the ammonium (NH<sub>4</sub><sup>+</sup>) form. Nitrogen fertilization increased shoot biomass but decreased root biomass with the highest shoot biomass occurring in the manure treatment. Compared to the no-N control, urea application did not affect shoot Cd concentration, but increased Cd content by 17 % due to shoot biomass increase. Chicken manure significantly decreased CaCl2-extractable Cd in soil, and the Cd concentration and total Cd uptake in the plant. Rhizosphere pH was the highest in the manure treatment and the lowest in the NH<sub>4</sub><sup>+</sup> treatments. The manure and nitrate (NO<sub>3</sub><sup>-</sup>) treatments tended to have

Responsible editor: Elena Maestri

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higher rhizosphere pH than their respective bulk soil pH, whereas the opposite was observed for urea and NH<sub>4</sub><sup>+</sup> treatments. Furthermore, the concentrations of extractable Cd in soil and Cd in the plant correlated negatively with rhizosphere pH. The study concludes that urea significantly enhanced the Cd phytoaccumulation by *C. rossii* while chicken manure decreased Cd availability in soil and thus the phytoextraction efficiency.

**Keywords** Ammonium (NH<sub>4</sub>–N) · Animal manure · Cd · Hyperaccumulator · Nitrate (NO<sub>3</sub>–N) · Phytoremediation

#### Introduction

Soil contamination by heavy metals has accelerated during the last few decades. Cadmium is considered as one of the most toxic heavy metals. The high concentrations of Cd observed in many agricultural soils are due to long-term use of phosphate fertilizer and sewage sludge (Stingu et al. 2012). As a non-essential element for living organisms, Cd is highly mobile in soil-plant systems and can adversely affect both human health and ecosystem functioning (Dwivedi et al. 2014). Therefore, cost-effective procedures are required to remediate Cd-contaminated soils.

Compared with classical remediation techniques, phytoremediation is a cheap and environmentally friendly technique and has attracted attention in recent years. Especially for agricultural soils, concentrations of heavy metals are generally low, and heavy metals are distributed on the surface and can be easily extracted by plant root systems. The task for effective phytoremediation is to identify hyperaccumulator plants with high biomass production. There are numerous reports on the use of hyperaccumulators to remove heavy metals from the soil (Bayramoglu et al. 2012;



Li et al. 2014). However, the slow growth and low shoot biomass limit the use of many hyperaccumulator plants for a large-scale remediation. Apart from the use of hyperaccumulators with larger biomass, the application of fertilizers or soil amendments to improve soil physicochemical properties and increase plant biomass production is also important for phytoremediation. For example, Wei et al. (2010) found that adding chicken manure significantly increased total Cd extraction by *Solanum nigrum* although it decreased Cd concentration in plant tissues.

Nitrogen (N) is an essential macronutrient and is involved in many metabolic processes including the biosynthesis of amino acids and proteins in plants. The application of N fertilizers is important for biomass production and yield. Nitrate (NO<sub>3</sub><sup>-</sup>) and ammonium (NH<sub>4</sub><sup>+</sup>) are the main inorganic N sources for plants in most agricultural systems. It has been reported how the supply of NH<sub>4</sub><sup>+</sup> enhances Zn and Cd uptake by non-hyperaccumulators sunflower and tobacco, which has been related to rhizosphere acidification (Loosemore et al. 2004; Zaccheo et al. 2006). Zaccheo et al. (2006) suggested that manipulation of rhizosphere pH via the combined use of NH<sub>4</sub><sup>+</sup>-N and nitrification inhibitors was an effective way to enhance phytoextraction of Cd and Zn from contaminated soils. However, in hydroponic experiments, supplying NO<sub>3</sub> doubled Cd concentration in the shoots of Noccaea caerulescens, compared with NH<sub>4</sub><sup>+</sup> supply (Xie et al. 2009). Similar results were found for Sedum plumbizincicola in solution culture; NO<sub>3</sub> stimulated Cd uptake by roots, xylem loading, shoot-to-root translocation, and uploading to the leaves and thus enhanced Cd phytoextraction (Hu et al. 2013). Monsant et al. (2008) showed that NO<sub>3</sub><sup>-</sup> nutrition improved Zn phytoextraction in solution culture by increasing the biomass and Zn concentration of shoots of N. caerulescens.

This soil-based pot experiment compared the effects of manure-N and different forms of chemical N supply (NO<sub>3</sub><sup>-</sup>–N, NH<sub>4</sub><sup>+</sup>–N, and urea-N) on Cd uptake by *Carpobrotus rossii*, a newly found Australian Cd hyperaccumulator (Zhang et al. 2014). A nitrification inhibitor was used to maintain the NH<sub>4</sub><sup>+</sup> form for plant uptake. We hypothesized that the NO<sub>3</sub><sup>-</sup>–N fertilizer form would increase Cd phytoextraction by *C. rossii* above that of NH<sub>4</sub><sup>+</sup>–N forms, in line with earlier research.

# Materials and methods

## **Cd-contaminated soil**

The soil was collected from the surface layer (0–15 cm) in the farm of La Trobe University (37° 43′ 18″ S, 145° 02′ 52″ E). Soil samples were air-dried and ground to pass through a 4-mm sieve. The soil had pH 4.99 (1:5, soil:water), cation

exchange capacity  $16 \text{ cmol kg}^{-1}$ , and the following properties (mg kg $^{-1}$ ): hydrolyzable N 158, available P 11.5, available K 132, and background Cd 0.5. The soil was spiked with 20 mg Cd kg $^{-1}$  by mixing CdCl $_2$  solution thoroughly into the soil. The water content was adjusted to 60 % field capacity. The spiked soil was placed in constant room temperature (25 °C) for 2 weeks to allow added Cd to equilibrate.

## Experimental design and growing conditions

The experiment consisted of six treatments with three replicates. The treatments were no-N control, chicken manure,  $Ca(NO_3)_2$ , urea  $[CO(NH_2)_2]$ ,  $(NH_4)_2SO_4$ , and  $(NH_4)_2SO_4$ plus NH<sub>2</sub>C(NH)<sub>2</sub>CN (dicyandiamide, DCD). The N and DCD were applied to the soil in 100 mL of water to bring the soil water content to 80 % field capacity. Nitrogen was applied at 20 mg N kg<sup>-1</sup> weekly (total=200 mg N kg<sup>-1</sup>). In the NH<sub>4</sub><sup>+</sup>+DCD treatment, DCD was applied at 8 mg kg<sup>-1</sup>. This application rate of DCD had no significant effect on plant growth and was effective in minimizing nitrification (Monsant et al. 2008). Chicken manure had the following properties: pH 7.68, total N 20.1 g kg<sup>-1</sup>, total Cd 0.55 mg kg<sup>-1</sup>, available N 3430 mg kg<sup>-1</sup>, available P 11.5 mg kg<sup>-1</sup>, and available K 150 mg kg<sup>-1</sup>. The manure was added to the soil at a rate of 1 % w/w (about 200 mg N kg<sup>-1</sup>) before planting.

A pot experiment was conducted in a glasshouse (20±5 °C) with natural light. Plastic pots were lined with plastic bags and filled with 1.5 kg air-dried soil. Basal nutrients were applied at the following concentration expressed in milligrams per kilogram of soil: 150 KNO<sub>3</sub>, 21 MgSO<sub>4</sub>·H<sub>2</sub>O, 150 KH<sub>2</sub>PO<sub>4</sub>, 236 CaCl<sub>2</sub>·2H<sub>2</sub>O, 18 MnCl<sub>2</sub>·4H<sub>2</sub>O, 0.67 H<sub>3</sub>BO<sub>3</sub>, 10.33 ZnSO<sub>4</sub>·7H<sub>2</sub>O, 1.42 CuCl<sub>2</sub>·5H<sub>2</sub>O, and 0.15 Na<sub>2</sub>MoO<sub>4</sub>·2H<sub>2</sub>O. The soil water content was maintained at around 80 % field capacity by watering daily for the experiment duration.

Pre-culture of stem cuttings of *C. rossii* (family Aizoaceae) were grown in plastic nursery cells for 3 weeks (Zhang et al. 2014), and then three uniform rooted stem cuttings were transplanted into each pot. The plants were harvested on day 78.

## Plant and soil analyses

At harvest, shoots were cut at their base, and roots were separated from the soil. Rhizosphere soils were collected by gently shaking off the soil adhering to the roots. The roots were first rinsed with running tap water. The root morphological parameters were determined with a root scanner at 600 dpi (Epson Perfection 4990 Scanner, model J131B, Epson Inc.). Both shoots and roots were then rinsed in distilled water, immersed in 0.1 mM HCl for approximate 5 s to remove external metal from the tissue surface, and finally rinsed with distilled water twice. These samples were dried in an oven at 105 °C



for 15 min and then at 70 °C until completely dry, weighed immediately upon removal from a desiccator, and then ground. Plant materials (0.5 g) were digested with 7 mL mixture of concentrated HNO<sub>3</sub> and HClO<sub>4</sub> (4:1 by volume).

The soil samples taken at the harvest were air-dried and ground to pass through 2 mm. For extractable Cd, air-dried soil was extracted in 0.01 M CaCl<sub>2</sub> solution  $(1:10\,\text{w/v})$  by shaking end-over-end for 16 h at 25 °C. The suspensions were then centrifuged at 3000 rpm for 10 min, followed by filtering the supernatant through Whatman No. 1 filter paper. The supernatant was used for Cd determination.

The concentrations of Cd in shoot and root digests, and soil extracts were determined using ICP-OES (Varian Vista AX CCD, Australia Pty Ltd.). All plant tissue concentrations were expressed on a dry weight basis. The pH of the bulk soil and rhizosphere soil was determined in the supernatant using a pH meter (Thermo Orion 720, USA) after shaking 5 g soil in 25 mL 0.01 M CaCl<sub>2</sub> solution, end-over-end for 12 h, and then centrifuging at 2000 rpm for 10 min.

## Statistical analysis

The data were statistically analyzed using Excel and SPSS 16.0 software, and expressed as means $\pm$ standard deviation. The means of the treatments were compared using the Duncan's multiple range tests at p=0.05.

#### Results

## Plant growth

All plants appeared healthy during the experiment, with differences in shoot growth between the treatments becoming noticeable prior to the harvest at 78 days. The average shoot dry weights ranged from 6.59 to 8.77 g pot<sup>-1</sup>. Relative to the no-N control, N fertilization increased shoot weight by 13–33 %. Shoot biomass was the highest in the manure treatment while DCD did not affect the shoot dry weight in the NH<sub>4</sub><sup>+</sup> treatments (Fig. 1a). In contrast, the application of manure and chemical N decreased the root dry weight by 38–47 % with manure having less effect (Fig. 1a), and hence, there was an increase in the shoot-to-root dry weight ratio. The shoot-to-root dry weight ratio was the highest for the urea and NH<sub>4</sub><sup>+</sup> treatments and the lowest for the no-N control with no significant differences between the four chemical N treatments (Fig. 1b).

## Root morphology

Quantitative data of the root morphology are given in Table 1. The decrease in root biomass with the chemical N treatments coincided with smaller surface areas and shorter cumulative

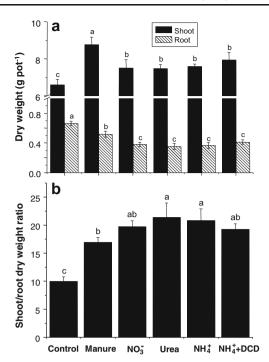


Fig. 1 Effects of N forms and nitrification inhibitor (DCD) on dry weights of (a) shoots and roots, and (b) shoot-to-root ratio of *C. rossii* grown in a Cd-contaminated soil for 78 days. The control had no added nitrogen. Treatments assigned with *different letters* are statistically different according to the Duncan's multiple range tests at the 5 % level. *Error bars*  $\pm$ standard deviation (n=3)

lengths of roots than the control plants. Furthermore, root length and root surface area were greater in the manure than in the chemical N treatments. Thicker roots with larger diameters occurred with the  $NO_3^-$ ,  $NH_4^+$ , and urea treatments.

#### Rhizosphere and bulk soil pH

The rhizosphere pH ranged from 4.57 to 5.65, while the bulk soil pH varied between 4.62 and 5.31 (Fig. 2). The bulk soil pH was highly correlated with the rhizosphere pH (p<0.01), indicating that the bulk soil had been affected by the root activity. Rhizosphere pH was the highest in the manure treatment and the lowest in the NH<sub>4</sub><sup>+</sup> treatments while the manure and NO<sub>3</sub><sup>-</sup> treatments tended to have higher rhizosphere pH and the urea and NH<sub>4</sub><sup>+</sup> treatments lower rhizosphere pH than their respective bulk soil pH readings.

## Cadmium in plants

The concentration of Cd in shoots varied from  $97 \text{ mg kg}^{-1}$  for the manure treatment to  $172 \text{ mg kg}^{-1}$  for the urea treatment (Fig. 3a). Compared to the no-N control, the manure and  $NO_3^-$  treatments had lower



**Table 1** Influence of different treatments on root phenotype of *C. rossii* 

Treatments	Root length (m plant <sup>-1</sup> )	Average diameter (µm)	Root surface (cm <sup>2</sup> plant <sup>-1</sup> )
Control	119±18a	178±3c	679±85a
Manure	110±9a	180±5c	627±45a
$NO_3^-$	77±5c	191±2a	466±30c
Urea	64±7d	191±8a	384±27d
$\mathrm{NH_4}^+$	82±12bc	186±3b	468±60c
NH <sub>4</sub> <sup>+</sup> +DCD	92±17b	180±2c	513±85b

Data are means $\pm$ standard deviation. Different letters indicate significant differences between treatments at p<0.05

while the urea and NH<sub>4</sub><sup>+</sup>+DCD treatments had similar concentrations of Cd in shoots.

The total amount of Cd in the shoots was significantly enhanced by supply of urea and  $\mathrm{NH_4}^+ + \mathrm{DCD}$ , but decreased by manure and  $\mathrm{NO_3}^-$  treatments, compared to the control (Fig. 3c). Thus, the Cd amount was the highest in the urea and  $\mathrm{NH_4}^+ + \mathrm{DCD}$  treatments, followed by the  $\mathrm{NH_4}^+$  treatment (similar to control) and then the manure and  $\mathrm{NO_3}^-$  treatments (lower than control). The amounts of Cd in the shoots in the urea and  $\mathrm{NH_4}^+ + \mathrm{DCD}$  treatments were 15 and 14 % greater than that in the control, respectively, and were greatest among the N treatments.

The concentrations of Cd in the roots were 4–6-fold higher than those in the shoots, ranging from 540 for the manure treatment to 793 mg kg $^{-1}$  for the urea treatment (Fig. 3a). DCD did not significantly affect the Cd concentration in roots although NH $_4$  $^+$ +DCD tended to have higher Cd than the NH $_4$  $^+$ -alone treatment (Fig. 3a, b). Unlike the shoots, the amount of Cd in the root did not vary between the N treatments, with Cd amount being lower in all N treatments compared with the control.

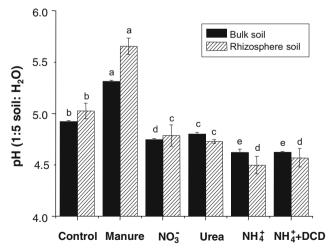


Fig. 2 Effects of N forms and manure addition on rhizosphere and bulk soil pH of C. rossii grown in a Cd-contaminated soil for 78 days. Means for the same type of soil with different letters are statistically different according to the Duncan's multiple range tests at the 5 % level. Error bars  $\pm$ standard deviation (n=3)

The shoot-to-root ratios of Cd concentration and content for the  $\mathrm{NH_4}^+$ ,  $\mathrm{NH_4}^+ + \mathrm{DCD}$  and urea treatments were significantly higher than those for the manure and  $\mathrm{NO_3}^-$  treatments, indicating that  $\mathrm{NH_4}^+$  and urea enhanced Cd translocation from the roots to shoots compared with that of  $\mathrm{NO_3}^-$  and manure (Fig. 3c, d). The highest shoot-to-root ratio of Cd concentration occurred in the control treatment.

Given the treatment effects on rhizosphere pH (Fig. 2) and Cd concentrations in shoots (Fig. 3a), it was thus not surprising that the Cd concentration in shoots correlated negatively with rhizosphere pH across treatments (p<0.01) (Fig. 4).

#### Extractable Cd in bulk soil

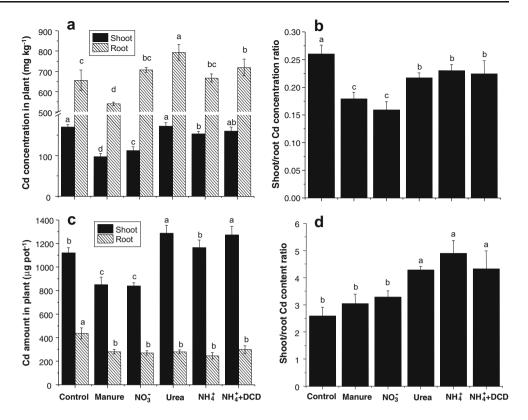
Compared to the control, most of N fertilizers tended to increase the concentration of CaCl<sub>2</sub>-extractable Cd but the only significant difference was recorded with the NH<sub>4</sub><sup>+</sup>+DCD treatment (Fig. 5). By contrast, manure amendment strongly decreased the concentration of CaCl<sub>2</sub>-extractable Cd by a factor of 2 compared to the control.

#### **Discussion**

This study demonstrated that C. rossii could accumulate Cd to a concentration of over 170 mg kg<sup>-1</sup> in the shoots, and that ammonium enhanced while nitrate decreased the Cd phytoextraction by this species when grown in the acid soil. C. rossii is a succulent glabrous halophytic perennial commonly found in the coastal fore-dunes of southern Australia (Venning 1988). It was shown to hyperaccumulate Cd and to be more tolerant to Cd than most species (Zhang et al. 2014). Other Cd hyperaccumulators have been identified, notably N. caerulescens (Robinson et al. 1998), Arabidopsis halleri (Bert et al. 2002), Rorippa globosa (Sun et al. 2007), S. nigrum (Wei et al. 2006), Sedum alfredii (Yang et al. 2004), Viola baoshanensis (Liu et al. 2004), and S. plumbizincicola (Li et al. 2014), but none of these plants are native to Australia (van der Ent et al. 2013). In combination with its easy growing, and salt- and drought-tolerant

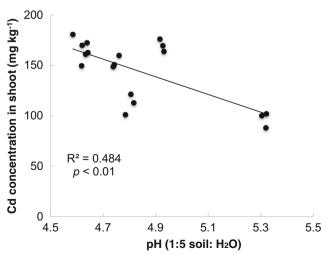


Fig. 3 Effects of N form and manure addition on (a) Cd concentrations in shoots and roots, (b) shoot-to-root Cd concentration ratio, (c) the amount of Cd in shoots and roots, and (d) shoot-to-root Cd content ratio of *C. rossii* grown in a Cd-contaminated soil for 78 days. Treatment means with *different letters* within a tissue type are statistically different according to the Duncan's multiple range tests at the 5 % level. *Error bars* ±standard deviation (n=3)



traits, this native species, *C. rossii*, could be a promising candidate for phytoremediation of Cd-contaminated soils, especially in drought-prone areas and soils with high salinity (Zhang et al. 2014).

Nitrogen fertilization is believed helpful for phytoremediation of heavy metals because it enhances plant growth, influences the cellular charge balance (Marschner and Römheld 1994) and helps to produce proteins which form complexes with metals to affect translocation and/or storage or are antioxidative enzymes to reduce the level of oxidation



**Fig. 4** Relationship between rhizosphere pH and Cd concentration in shoots for *C. rossii* grown in a Cd-contaminated soil for 78 days

caused by metals (Sarwar et al. 2010). Nitrogen fertilizers, usually in the form of  $\mathrm{NH_4}^+$ ,  $\mathrm{NO_3}^-$ , urea, or organic nitrogen (manure), are used to enhance plant biomass and the phytoextraction efficiency of contaminants from soils (Giansoldati et al. 2012). This study showed that all forms of nitrogen could efficiently increase plant biomass to a similar extent. Although the highest Cd concentrations in shoots were observed in the control, urea, and  $\mathrm{NH_4}^+$ +DCD treatments, the highest amounts of Cd were obtained in the urea and  $\mathrm{NH_4}^+$ +

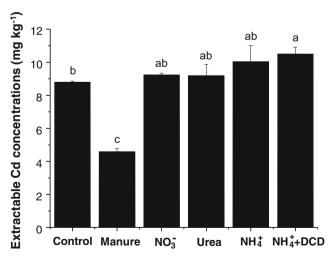


Fig. 5 Concentrations of  $CaCl_2$ -extractable Cd in bulk soil receiving different treatments. Means with *different letters* are statistically different according to the Duncan's multiple range tests at the 5 % level. *Error bars* standard deviation (n=3)



DCD treatments due to higher biomass production in these treatments compared to the control (Fig. 3).

Nitrate and  $NH_4^+$  are the two main forms of inorganic N that are taken up by plants. Since N accounts for 80 % of the total nutrients taken up by roots, the charge of these N ions has a strong effect on charge balance in root epidermal cells (Marschner and Römheld 1994). Physiologically, the uptake of NO<sub>3</sub> and NH<sub>4</sub> induces the release of OH /HCO<sub>3</sub> and H ions, respectively. This was shown in the present study where the rhizosphere pH was lower in the NH<sub>4</sub><sup>+</sup> than in the NO<sub>3</sub><sup>-</sup> treatment (Fig. 2). Ammonium nutrition causes rhizosphere acidification which in turn increases metal bioavailability in soil and hence plant uptake (Zaccheo et al. 2006). Thus, rhizosphere acidification might explain the higher concentration and phytoextraction of Cd by C. rossii grown in the NH<sub>4</sub><sup>+</sup> than the NO<sub>3</sub><sup>-</sup> treatment, with much lower rhizosphere pH in the NH<sub>4</sub><sup>+</sup> treatments. This is also supported by the strong negative relationship between Cd concentration in shoots and rhizosphere pH (Fig. 4), and the higher Cd phytoextraction with the NH<sub>4</sub><sup>+</sup>+DCD than the NH<sub>4</sub><sup>+</sup>-alone treatment (Fig. 3) because DCD is able to inhibit nitrification and thereby decrease NO<sub>3</sub><sup>-</sup> concentration in soil (Arnamwong et al. 2015). Our results are consistent with the study of Zaccheo et al (2006) who showed that the application of NH<sub>4</sub><sup>+</sup> together with nitrification inhibitor DMPP resulted in higher Cd uptake compared to NO<sub>3</sub>, in the shoot of non-hyperaccumulator sunflower grown in contaminated soils.

The present study showed that manure addition increased the production of shoot biomass but decreased the concentration of extractable Cd in soil and Cd concentration in plants. A number of studies have also shown that the application of organic materials (e.g., farmyard manure, compost, peat soil) resulted in a decrease in Cd concentration in plants (e.g., corn, wheat, radish, alfalfa, and Indian mustard) (Wei et al. 2010). In general, organic amendments such as animal manures contain a high proportion of humidified organic matter, which forms strong complexes with Cd, resulting in Cd immobilization and decreased Cd bioavailability in the soil (Putwattana et al. 2010). Additionally, manures often contain high contents of dissolved organic matter which forms soluble organic-metal complexes (Bolan et al. 2003a; Vaca-Paulin et al. 2006) that reduce metal availability to plants (Bolan et al. 2003b). Soluble organic-Cd complexes are also likely to have contributed to the lower phytoextraction by the manure treatment.

All N fertilizers reduced root biomass (Fig. 1a), and this is likely to be in response to the alleviation of N deficiency in the soil. The chemical N treatments resulted generally in shorter, thicker roots with less surface area compared with the control and manure treatments (Table 1). An increased biomass allocation to the roots is a common phenomenon in many plant species in response to N deficiency. It is suggested that N deficiency initiates transcriptional changes that lead to the

accumulation of carbohydrates in the shoots and increase the translocation of sucrose to the roots (Hermans et al. 2006). It could be argued that small root systems in the N-amended treatments would not favor the uptake of heavy metals in soil. For example, Lombi et al. (2000) reported the greater root system in the French ecotype of *N. caerulescens* (root:shoot=0.21) than the Prayon ecotype (root:shoot=0.12) could be partly responsible for the higher Cd uptake in roots and accumulation in shoots of the French ecotype. However, in this study, the NH<sub>4</sub><sup>+</sup>+DCD and urea treatments resulted in higher Cd uptake by shoots compared to the control and yet had significantly smaller root systems. Thus, soil factors such as rhizosphere pH and Cd extractability were more important in determining Cd extraction from the soil than root morphology.

Nitrogen form also appeared to affect Cd translocation from roots to shoots of C. rossii. It is evident that the shootto-root ratio of Cd concentration and content was significantly higher in the urea and NH<sub>4</sub><sup>+</sup> treatments than the NO<sub>3</sub><sup>-</sup> treatment (Fig. 3c, d), indicating that urea and NH<sub>4</sub><sup>+</sup> nutrition stimulates the translocation of Cd from roots to shoots to a higher extent compared with NO<sub>3</sub> nutrition. Wang et al. (2010) also found the shoot-to-root Cd concentration in mustard grown in pots was higher with NH<sub>4</sub><sup>+</sup> than with NO<sub>3</sub><sup>-</sup> treatment. The mechanism by which N form affects the Cd translocation is unclear. In N. caerulescens, NO<sub>3</sub> nutrition enhanced the formation of organic acids while NH<sub>4</sub><sup>+</sup> nutrition increased the formation of amino acids (White-Monsant and Tang 2013). The form of N in this study might have also affected the concentrations of different metabolites in the xylem of C. rossii and thereby affected the Cd transport from the roots to the shoots. However, N form did not affect Zn speciation in shoots, roots, or xylem even though Zn accumulation in the shoots was enhanced (Monsant et al. 2008, 2011).

The results from this study, where NH<sub>4</sub><sup>+</sup>-N forms resulted in greater Cd uptake by shoots than NO<sub>3</sub><sup>-</sup>-N, mean that our hypothesis is rejected. The results are inconsistent with those of Xie et al. (2009) and Hu et al. (2013). By using a positronemitting <sup>107</sup>Cd tracer, Hu et al. (2013) showed that Cd transport was 1.5-fold and 3.7-fold faster in the roots and shoots, respectively, in the Cd hyperaccumulator S. plumbizincicola supplied with NO<sub>3</sub><sup>-</sup> than with NH<sub>4</sub><sup>+</sup>. The concentration of Cd was 2.63 times higher in shoots of the NO<sub>3</sub><sup>-</sup>- than NH<sub>4</sub><sup>+</sup>-fed plants in nutrient solution. Similarly, Xie et al. (2009) showed in hydroponic experiments that compared with NH<sub>4</sub><sup>+</sup> nutrition, NO<sub>3</sub> nutrition doubled shoot Cd concentrations and increased shoot Cd accumulation by 2.6-8-fold in N. caerulescens. The reasons for the discrepancy in these studies can be attributed to the rhizosphere effect in this soilbased study. This effect, where the rhizosphere pH differs with N form (Fig. 2), is dissipated in solution culture, as ions can freely diffuse away from the root surface in the culture solution, which cannot happen in the soil. Thus, the study



highlights the importance of soil-based experiments, compared to those using solution culture, when investigating Cd phytoextraction from Cd-contaminated soils.

#### **Conclusions**

This study aimed at testing chemical N fertilizers and manure to evaluate the influence of N form on Cd phytoextraction using Australian native species C. rossii, a Cd hyperaccumulator. The addition of N fertilizers significantly enhanced shoot biomass production, particularly manure application resulted in the highest shoot biomass. However, manure application significantly decreased Cd availability in soil and hence plant Cd phytoextraction efficiency. Although the four chemical N treatments resulted in similar shoot and root biomass production, the highest efficiency of Cd phytoextraction by C. rossii was in the urea- and NH<sub>4</sub><sup>+</sup>+ DCD-amended soils. Thus, this study suggests that urea would be a good N source to enhance Cd phytoextraction using C. rossii. Future work is required to compare the effect of N form on metal extraction with other plant species and to validate the efficiency of *C. rossii* under field conditions.

**Acknowledgments** We thank the Key Projects in the National Science & Technology Pillar Program (2015BAD05B04), National High Technology Research and Development Program of China (863 Program) (2012AA06A204), Jiangsu Provincial Natural Science Foundation of China (BK2012891), and Australian Research Council Linkage Project (LP100100800) for the financial support. We also thank anonymous reviewers for their valuable comments.

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