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## COMMUNICATION

## A highly selective turn-on near-infrared fluorescent probe for hydrogen sulfide detection and imaging in living cells†

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**We have described a turn-on near-infrared fluorescent probe Cy–NO<sub>2</sub> based on nitro group reduction for intracellular H<sub>2</sub>S detection. The probe employs cyanine dye as a fluorophore, and is equipped with a nitro group as a fluorescent modulator. It is readily employed for assessing intracellular H<sub>2</sub>S level changes, and confocal imaging is achieved successfully.**

Hydrogen sulfide (H<sub>2</sub>S) is well known for its unpleasant rotten egg smell, which was considered as a toxic gas for a long time. However, many studies on endogenous H<sub>2</sub>S have demonstrated that hydrogen sulfide is the most recent endogenous gasotransmitter along with nitric oxide (NO) and carbon monoxide (CO).<sup>1</sup> Endogenous hydrogen sulfide is biosynthesized by at least three separate enzymes: cystathionine-β-synthase (CBS),<sup>2</sup> cystathionine-γ-lyase (CSE),<sup>3</sup> and 3-mercaptopyruvate sulfurtransferase (3-MST).<sup>4</sup> It is reported that mitochondrial sulfide quinone oxidoreductase (SQR) and persulfide dioxygenase (SDO) are involved in the consumption of H<sub>2</sub>S.<sup>5</sup> H<sub>2</sub>S also appears to regulate inflammation.<sup>6</sup> Furthermore, some research studies also have indicated that the hydrogen sulfide levels are related to Down syndrome and Alzheimer's disease.<sup>7</sup>

The interest in understanding the physiological and pathological functions of hydrogen sulfide has been increasing steadily. A number of sensitive and selective methods have been developed for the detection of H<sub>2</sub>S including colorimetry,<sup>8</sup> electrochemical analysis,<sup>9</sup> and gas chromatography.<sup>10</sup> However, these techniques often require post-mortem processing and/or destruction of tissues or cells,<sup>11</sup> and are hence not suitable for the analysis of endogenous H<sub>2</sub>S in real-time. Compared with these biological detection technologies, the fluorescence method provides greater sensitivity, less invasiveness, and more convenience, and offers high sensitivity as well as real-time imaging.<sup>12</sup> Recently, there has been an explosive increase in the number of fluorescent probes for H<sub>2</sub>S.<sup>13</sup> In general, these probes are divided into three types: exploiting a unique reduction reaction between an azide group and H<sub>2</sub>S;<sup>13a–c,15c</sup> taking advantage of the double nucleophilic character of H<sub>2</sub>S; utilizing a copper-centered coordination

complex in which Cu<sup>2+</sup> can be released by binding H<sub>2</sub>S. However, most of these fluorescent probes emit in the ultra-violet or the visible region, which can be interfered by cell auto-fluorescence easily. In contrast, long wavelength probes with emission in the near-infrared (NIR) region are optimal for biological imaging applications due to minimal photo damage to biological samples and minimum interference from background auto-fluorescence in living systems.<sup>14</sup> Therefore, it is necessary to develop new types of fluorescent probes that can be used for H<sub>2</sub>S detection under physiological conditions, preferably with emission located in the NIR region. On the other hand, it is reported that the nitro group can be reduced by Na<sub>2</sub>S to produce the corresponding amino group under mild conditions,<sup>15</sup> which inspires us to design and synthesize a new type of fluorescent probe containing a nitro group for H<sub>2</sub>S detection in cells. Herein, we present a new NIR fluorescent probe Cy–NO<sub>2</sub> for detection of H<sub>2</sub>S in living cells.

As an overall strategy, the heptamethine cyanine was particularly chosen as a fluorophore, because of its high molar absorption coefficient and NIR emission, along with other members of the cyanine family, which have been used for a variety of fluorescence sensing applications. It is speculated that the fluorescence properties of the heptamethine cyanine dye could be modulated *via* a photo-induced electron transfer (PET) process from the excited fluorophore to a strong electron-withdrawing group (donor-excited PET; d-PET).<sup>16</sup> Therefore, *m*-nitrophenol was selected as a modulator for the Cy–NO<sub>2</sub> probe. After being equipped with *m*-nitrophenol, the fluorescence of the cyanine platform was quenched by the electron transfer process between the modulator and the fluorophore. As expected, the nitro group could be reduced to the amino group under physiological conditions, triggering an increase in the fluorescence emission, thereby allowing the formation of a “turn-on” fluorescent probe for H<sub>2</sub>S detection. The structure and the proposed mechanism of fluorescence probe Cy–NO<sub>2</sub> for H<sub>2</sub>S detection are shown in Scheme 1.

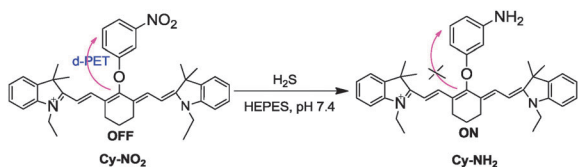
To verify whether the probe was suitable for the physiological detection, we evaluated the effect of pH on the fluorescence of the probe.<sup>17</sup> As shown in Fig. S1 (ESI†), the pH of the medium had hardly any effect on the fluorescence of Cy–NO<sub>2</sub> and its reduced product Cy–NH<sub>2</sub> over the pH range of 4.2–8.2. Therefore, the probe was expected to work well under physiological conditions (40 mM HEPES, pH 7.4).

The absorption and fluorescence spectra of the probe were examined under simulated physiological conditions (40 mM HEPES, pH 7.4, 10 μM Cy–NO<sub>2</sub>). In the absence of H<sub>2</sub>S,

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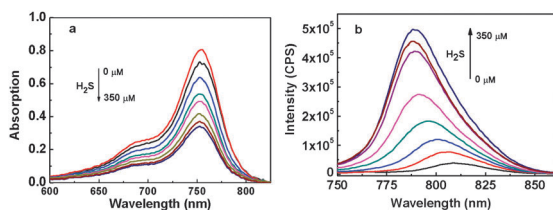


**Scheme 1** Structures of Cy-NO<sub>2</sub> and its reduced product Cy-NH<sub>2</sub>, and the proposed mechanism of fluorescence probe Cy-NO<sub>2</sub> for H<sub>2</sub>S detection.

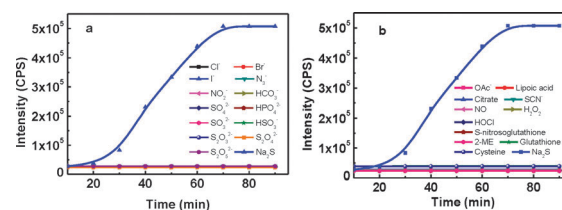
Cy-NO<sub>2</sub> exhibited  $\lambda_{\text{max}}$  for absorption and emission at 755 nm and 809 nm, respectively, both of which lie in the NIR region (Fig. 1). There is a significant decrease in the absorption spectrum upon addition of H<sub>2</sub>S (Fig. 1a). We attributed the phenomenon of absorption decrease to the difference of molar absorptivity between Cy-NO<sub>2</sub> and Cy-NH<sub>2</sub>. We calculated that the molar absorptivity of Cy-NH<sub>2</sub> ( $\epsilon_{755\text{nm}} = 3.4 \times 10^4 \text{ M}^{-1} \text{ cm}^{-1}$ ) was much lower than that of Cy-NO<sub>2</sub> ( $\epsilon_{755\text{nm}} = 8.1 \times 10^4 \text{ M}^{-1} \text{ cm}^{-1}$ ). Therefore, the absorption spectrum shows a significant decrease after addition of H<sub>2</sub>S. Upon addition of different concentrations of H<sub>2</sub>S to the buffer solution which contained 10  $\mu\text{M}$  Cy-NO<sub>2</sub>, the fluorescence intensity increased by  $\sim 12.7$  fold and the quantum yield increased from 0.05 to 0.11. Interestingly, the emission spectrum showed little blue shift in  $\lambda_{\text{max}}$  from 809 nm to 789 nm (Fig. 1b). This phenomenon was induced by the push-pull electronic effect of the functional moiety.<sup>18</sup> Upon reduction by H<sub>2</sub>S, the probe contained an electron-donating group (Ph-NH<sub>2</sub> moiety) instead of an electron-withdrawing group (Ph-NO<sub>2</sub> moiety). Therefore, the fluorescence spectrum underwent a slight blue shift.

During the experiments, interestingly, we found that the reaction rate between Cy-NO<sub>2</sub> and H<sub>2</sub>S was severely affected by the reaction temperature. In order to optimize the reaction temperature, we tested the effect of temperature on the reaction rate. The probe (10  $\mu\text{M}$ ) and 350  $\mu\text{M}$  Na<sub>2</sub>S were equilibrated at different temperatures, and the fluorescence intensity was acquired in 40 mM HEPES (pH 7.4) with emission at 789 nm. As shown in Fig. S2 (ESI†), the reaction was very slow at room temperature. The fluorescence intensity could reach saturation within 40 min at 60 °C, but the temperature was too high to detect H<sub>2</sub>S in living cells. Excitedly, the probe exhibited almost the same reaction rate under 37 °C and 45 °C conditions. The fluorescence intensity could reach saturation within 60 min. Thus, 37 °C was chosen as the optimal reaction temperature.

To verify whether there is fluorescence response to other biological analytes or not, Cy-NO<sub>2</sub> (10  $\mu\text{M}$ ) was treated with various biologically relevant analytes in HEPES buffer (40 mM, pH 7.4). Overall 24 biological species were screened. As shown in Fig. 2, Cy-NO<sub>2</sub> showed selective response for



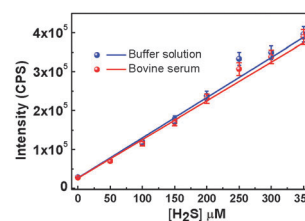
**Fig. 1** (a) Absorption and (b) fluorescence spectra of probe Cy-NO<sub>2</sub> (10  $\mu\text{M}$ ) treated with various concentrations of H<sub>2</sub>S: 0, 50, 100, 150, 200, 250, 300, 350  $\mu\text{M}$ . Spectra were acquired in 40 mM HEPES, pH 7.4, at 37 °C after incubation of the probe with H<sub>2</sub>S for 60 min with excitation at 755 nm and emission ranging from 755 to 900 nm.



**Fig. 2** Fluorescence responses and time courses of 10  $\mu\text{M}$  Cy-NO<sub>2</sub> to biologically relevant species for 90 min. (a) Anions and RSS. (b) RNS and ROS. Data shown: Na<sub>2</sub>S, HSO<sub>3</sub><sup>-</sup>, SO<sub>3</sub><sup>2-</sup>, S<sub>2</sub>O<sub>3</sub><sup>2-</sup>, S<sub>2</sub>O<sub>4</sub><sup>2-</sup>, S<sub>2</sub>O<sub>5</sub><sup>2-</sup>, NO, H<sub>2</sub>O<sub>2</sub> at 300  $\mu\text{M}$ . Cl<sup>-</sup>, Br<sup>-</sup>, I<sup>-</sup>, N<sub>3</sub><sup>-</sup>, NO<sub>2</sub><sup>-</sup>, OAc<sup>-</sup>, lipoic acid, citrate, glutathione, cysteine, S-nitrosoglutathione and 2-mercaptoethanol (2-ME) at 1 mM.

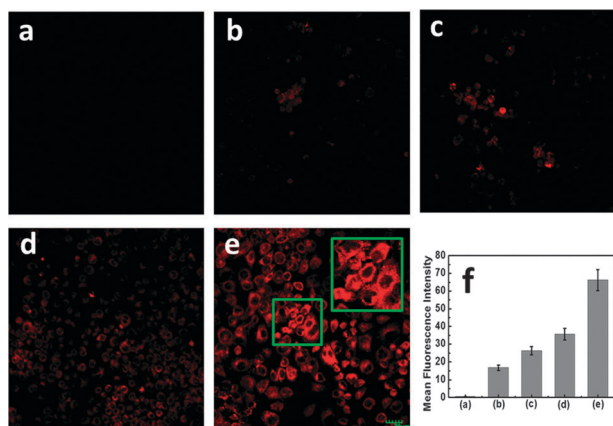
H<sub>2</sub>S over reactive oxygen species (ROS), reactive nitrogen species (RNS) and anions. Of all the tested reactive sulfide species (RSS), only S<sub>2</sub>O<sub>3</sub><sup>2-</sup>, S<sub>2</sub>O<sub>4</sub><sup>2-</sup>, S<sub>2</sub>O<sub>5</sub><sup>2-</sup>, glutathione and cysteine gave a limited increase in the fluorescence intensity. However, the intensity of the fluorescence increase was far weaker than that caused by H<sub>2</sub>S. To check whether the probe might turn on fluorescence upon incubation with other biological species over time, the probe's time courses with various species for 90 min were measured. As demonstrated in Fig. 2, Cy-NO<sub>2</sub> can selectively respond to H<sub>2</sub>S and avoid the interference of other relevant species. The experimental results displayed that the selective reduction of Cy-NO<sub>2</sub> to Cy-NH<sub>2</sub> could be used for the detection of H<sub>2</sub>S under simulated physiological conditions.

We explored the ability of the probe to quantify H<sub>2</sub>S in buffer solution and fetal bovine serum sample. To evaluate the ability of Cy-NO<sub>2</sub> in the detection of H<sub>2</sub>S concentration, the probe was treated with different concentrations of H<sub>2</sub>S (0–350  $\mu\text{M}$ ). The final concentration of the probe was maintained at 10  $\mu\text{M}$ . For the purpose of accurate analysis, a linear relationship is always necessary. Herein, we obtained a calibration curve between fluorescence emission intensity and H<sub>2</sub>S concentration. As shown in Fig. 3, the fluorescence signal was linearly related to the concentration of H<sub>2</sub>S in the given concentration range. The regression equation was  $F_{789\text{nm}} = 1111 \times [\text{H}_2\text{S}] + 19\,632$ , with  $r = 0.992$ . Next, we used fetal bovine serum to investigate whether our probe could detect H<sub>2</sub>S in complex biological samples. We prepared the fetal bovine serum samples containing H<sub>2</sub>S in different concentrations (0–350  $\mu\text{M}$ ). Following the above method, we got another calibration curve between fluorescence emission intensity and H<sub>2</sub>S concentration. The regression equation was  $F_{789\text{nm}} = 1073 \times [\text{H}_2\text{S}] + 18\,012$ , with  $r = 0.996$ . These results indicated that Cy-NO<sub>2</sub> could



**Fig. 3** The relationship between fluorescence intensity and H<sub>2</sub>S concentration in buffer solution (40 mM HEPES, pH 7.4) (blue) and commercial fetal bovine serum (red). The concentration of fetal bovine serum in the solution is 15% (v/v). 10  $\mu\text{M}$  Cy-NO<sub>2</sub> and H<sub>2</sub>S 0–350  $\mu\text{M}$ . The fluorescence intensity was acquired at 37 °C after incubation of the probe with H<sub>2</sub>S for 60 min ( $\lambda_{\text{ex}} = 755 \text{ nm}$ ,  $\lambda_{\text{em}} = 789 \text{ nm}$ ).





**Fig. 4** Confocal fluorescence images of living RAW264.7 cells incubated with various concentrations of Na<sub>2</sub>S. RAW264.7 cells loaded with 10 μM Cy-NO<sub>2</sub> and Na<sub>2</sub>S for 30 min. (a) Control, (b) 50 μM, (c) 150 μM, (d) 250 μM, and (e) 350 μM Na<sub>2</sub>S. Scale bar is 20 μm. Incubation was performed at 37 °C under a humidified atmosphere containing 5% CO<sub>2</sub>. The fluorescence was collected at 650–800 nm upon excitation at 635 nm. (f) The relationship between average fluorescence intensity and added various Na<sub>2</sub>S concentrations in a–e correspondingly. The cell body regions in the visual field (a–e) were selected as the regions of interest (ROI), and the average fluorescence intensity was determined *via* confocal laser-scanning microscopy. Data were normalized to controls and statistical analyses were performed with a two-tailed Student's *t*-test. \**P* < 0.05 (*n* = 3). Error bars are ± s.e.m.

qualitatively and quantitatively detect H<sub>2</sub>S in complex biological systems.

Having demonstrated the selectivity and sensitivity of Cy-NO<sub>2</sub> for H<sub>2</sub>S *in vitro*, we next established the ability of Cy-NO<sub>2</sub> to track H<sub>2</sub>S level changes in living cells by using a RAW264.7 cell model. RAW264.7 cells were incubated with 10 μM Cy-NO<sub>2</sub> for 10 min at 37 °C, and then washed with physiological saline to remove excess Cy-NO<sub>2</sub>. The treated cells were incubated with buffer containing different concentrations of Na<sub>2</sub>S (50, 150, 250 and 350 μM). After incubation for 30 min in RPMI 1640 Medium at 37 °C, the cells were washed with physiological saline to remove the excess Na<sub>2</sub>S, and then the cells were imaged by a confocal fluorescence microscope. The fluorescence was collected at 650–800 nm upon excitation at 635 nm. As a control, the cells not treated with Na<sub>2</sub>S were also imaged. The control experiments showed faint fluorescence (Fig. 4a), but those treated with various concentrations of Na<sub>2</sub>S displayed different fluorescence intensities. The confocal fluorescence images grew brighter as the concentrations of Na<sub>2</sub>S increased from 50 to 350 μM (Fig. 4b–e). The cell body regions in the visual field (Fig. 4a–e) were selected as the regions of interest (ROI), and the average fluorescence intensity was determined *via* confocal laser-scanning microscopy with various H<sub>2</sub>S concentrations (Fig. 4f). The results suggested that Cy-NO<sub>2</sub> had good membrane permeability, and these data also established that Cy-NO<sub>2</sub> could respond to intracellular H<sub>2</sub>S level changes within living cells.

We also applied Cy-NO<sub>2</sub> to the subcellular locations of H<sub>2</sub>S in the RAW264.7 cells using confocal fluorescence microscopy. The cells, with the same condition used in Fig. 4e, were co-stained with Cy-NO<sub>2</sub> (10 μM) and Janus Green B (JGB, 1 μM) for 15 min. Fig. S5 (ESI†) further reveals the location of the probe in the cytoplasm of these living RAW264.7 cells. We also employed

the Pearson correlation coefficient (*r*) which was used to quantify the degree of colocalization between fluorophores<sup>19</sup> to further reveal the subcellular locations of Cy-NO<sub>2</sub>. By using Olympus software, we obtained the value of Cy-NO<sub>2</sub> with JGB *r* = 0.85, revealing that Cy-NO<sub>2</sub> primarily locates in the cytoplasm.

In summary, we have developed a new NIR fluorescent probe that exhibits high selectivity and sensitivity for H<sub>2</sub>S both in aqueous solution and living cells. Our probe Cy-NO<sub>2</sub> shows remarkable turn-on fluorescence for H<sub>2</sub>S compared to other biologically relevant species. Confocal microscopy images indicate that our probe can detect the level changes of H<sub>2</sub>S in living cells. We anticipate that the fluorescent probe will be of great benefit for biomedical researchers to investigate the effects of H<sub>2</sub>S in biological systems.

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