



Modeling primary and secondary fractionation effects and atmospheric transport of polychlorinated biphenyls through single-source emissions

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Abstract The Chinese Gridded Industrial Pollutants Emission and Residue Model (ChnGIPERM) was used to investigate potential fractionation effects and atmospheric transport of polychlorinated biphenyls (PCBs) derived from single-source emissions in China. Modeling the indicative PCBs (CB28, CB101, CB153, and CB180) revealed spatiotemporal trends in atmospheric transport, gas/particle partitioning, and primary and secondary fractionation effects. These included the inference that the Westerlies and East Asian monsoons affect atmospheric transport patterns of PCBs by influencing the atmospheric transport time (ATT). In this study, dispersion pathways with long ATTs in winter tended to have short

ones in summer and vice versa. The modeled partitioning of PCB congeners between gas and particles was mainly controlled by temperature, which can further influence the ATT. The potential for primary and secondary fractionation was explored by means of numerical simulations with single-source emissions. Within ChnGIPERM, these phenomena were mainly controlled by the temperature and soil organic carbon content. The secondary fractionation of PCBs is a slow process, with model results suggesting a timescale of several decades.

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Introduction

Polychlorinated biphenyls (PCBs) are one of the “dirty dozen” (Stockholm Convention) or “legacy POPs” (persistent organic pollutants) due to their high toxicity, persistence, high bioaccumulation through the food chain, and propensity for long-range atmospheric transport (LRAT) (UNEP 2001). Once emitted, POPs can be globally dispersed through processes including LRAT and ocean currents and accumulate in environmental media such as soil, water, sediment, coast, and biota (Wania and Mackay 1993, 1996; Wania et al. 1999; Xing et al. 2005; Ren et al. 2007; Yadav et al. 2017; Kim et al. 2017; Pokhrel et al. 2018; Hanh et al. 2018). Since the 1930s, PCBs have been used in many commercial products. Due to their thermal stability, extraordinary dielectric properties, and oxidation resistance, PCBs were mostly applied in products such as transformers and capacitors (Bozlake et al. 2008). However, PCBs have also been associated with a number of undesirable health outcomes leading to their production being banned by the Stockholm Convention in 2001 (Wu et al. 2011; Wang et al. 2017; Cappelletti et al. 2018; Coxon et al. 2018).

To describe the atmospheric transfer of DDT from continents to oceans, the term “global distillation” was coined by Goldberg (1975). Subsequently, after entering the environment, the redistribution process of a pollutant may be latitudinally fractionated, depending on the ambient temperature and volatility of the pollutant. This has been termed “global fractionation” (Wania and Mackay 1993). In the redistribution process, however, LRAT acts as a “carrier” and moves chemicals from warmer regions to colder regions. Coincidentally, in many cases, the transfer of pollutants is from emission sources to more remote or pristine areas (Meijer et al. 2003; von Waldow et al. 2010). This movement is accompanied by dry and wet deposition and has been recognized as the most important pathway for PCBs and other POPs (Wania and Mackay 1996). At the same time, LRAT is a compelling problem in the study of environmental

distribution and fate due to elevated concentrations of POPs in polar regions and the concomitant “fractionation effect” theory. This quandary has led to numerous research studies attempting to quantify the global or regional transport and fate of POPs using measured or modeled data (Lohmann et al. 2007). Analogous fractionation of PCBs has been reported in several studies. For example, “urban fractionation” of PCBs was reported by Harner et al. (2004) for a relatively short distance (~ 75 km) along an urban–rural transect in Toronto. While their study did not reference the urban fractionation effect, Motelay-Massei et al. (2004) observed a clear gradient of one or two orders of magnitude for PAHs and PCBs as they moved from industrial to remote sites in the Seine River basin in France. According to soil monitoring data across China, Ren et al. (2007) observed PCB fractionation in soil from an urban–rural transect in the city of Shanghai (~ 200 km). “Urban fractionation” is less associated with “grasshopping” and air–soil exchange and more the result of proximity to a source region (i.e., a primary deposition area) (Li et al. 2010). In addition, “longitudinal fractionation” was observed by Ren et al. (2007) in a rural Chinese region due to most PCB usage in the eastern region than the western region in China. The latitudinal fractionation effect was also reported by Meijer et al. (2002, 2003), and their measured data included both soil and air data from southern UK to northern Norway.

Since the primary and secondary fractionation effects were proposed by Li et al. (2010), many scientists have shown a strong interest in studying the environmental distribution and behavior of POPs. For instance, primary sources (emission) influenced the distribution and background concentrations of PCBs in soils in the UK and Norway between 1998 and 2008 (Schuster et al. 2011). A primary fractionation effect was observed in the Tianjin region of China, with levels of PCBs dependent on proximity to major emission sources (Li et al. 2011). However, these phenomena are based on measured air or soil data, and only a few studies have modeled the fractionation effect (Sweetman et al. 2002; Gouin et al. 2004; Scheringer et al. 2004; von Waldow et al. 2010; Lammel and Stemmler 2012; Stemmler and Lammel 2012; Zhao et al. 2017) or mountainous cold condensation effects using the ClimoChem model (Wegmann et al. 2006). This limits our ability to evaluate PCB pollution in a strategic way, limits our ability to make

predictions about important sinks for PCBs, and ultimately reduces our ability to use limited resources more effectively.

Multimedia environmental models are useful tools for estimating fractionation phenomena (Mackay 2001). These models are able to incorporate physicochemical properties of individual PCBs, the geochemical properties of the soil and meteorological conditions; all of which can affect the distribution characteristics and atmospheric transport of PCBs. For example, China has a westerly and monsoon climate in the east part of the country. Additionally, the relatively complicated fractionation phenomena are difficult to observe by monitoring methods because they are influenced by many factors (meteorological conditions and soil properties), and the results are limited by sampling costs, compound pollution from different emission sources, as well as human activities or bioturbation.

In this study, we use numerical simulations using the Chinese Gridded Industrial Pollutants Emission and Residue Model (ChnGIPERM) (Cui et al. 2017) to explore the environmental behavior of PCBs derived from single-source emissions. The major objectives of this study are (1) to assess the atmospheric transport of PCBs derived from single-source emissions; (2) to determine the major factors that can influence the gas-particle partitioning of different PCB congeners in different regions or receptor sites, and (3) to explore primary and secondary fractionation.

Methodology

Model description

ChnGIPERM is a gridded mass balance model using a gridded system with a $1/6^\circ$ latitude and $1/4^\circ$ longitude resolution. A detailed description of the model can be found in Cui et al. (2017 & SI). Briefly, the main model platform includes both transfer and transport modules.

For the transfer module, a level IV fugacity method is applied within each grid cell in order to calculate and describe temporal changes to concentrations of industrial pollutants as well as intercompartmental transfer of the modeled chemicals within a multimedia system. The transfer module consists of six soil types (urban land, dry cropland, paddy field, forestry,

grassland, and uncultured land) with four soil vertical layers (or horizons), as well as water, sediment, and air (comprised of an atmospheric boundary layer (ABL, 0–1000 m) and an atmospheric low troposphere (ALT, 1000–4000 m)) compartments.

The transport module describes the mass exchange of chemicals between grid cells, which is driven by atmospheric transport (wind direction and speed) and water currents (Cui et al. 2017 and SI). The ChnGIPERM domain covers all of China spanning from 17° to 55°N and 70° to 135°E with 228 latitudinal grids and 260 longitudinal grids (Fig. 1).

Model experimental design and physicochemical properties of PCBs

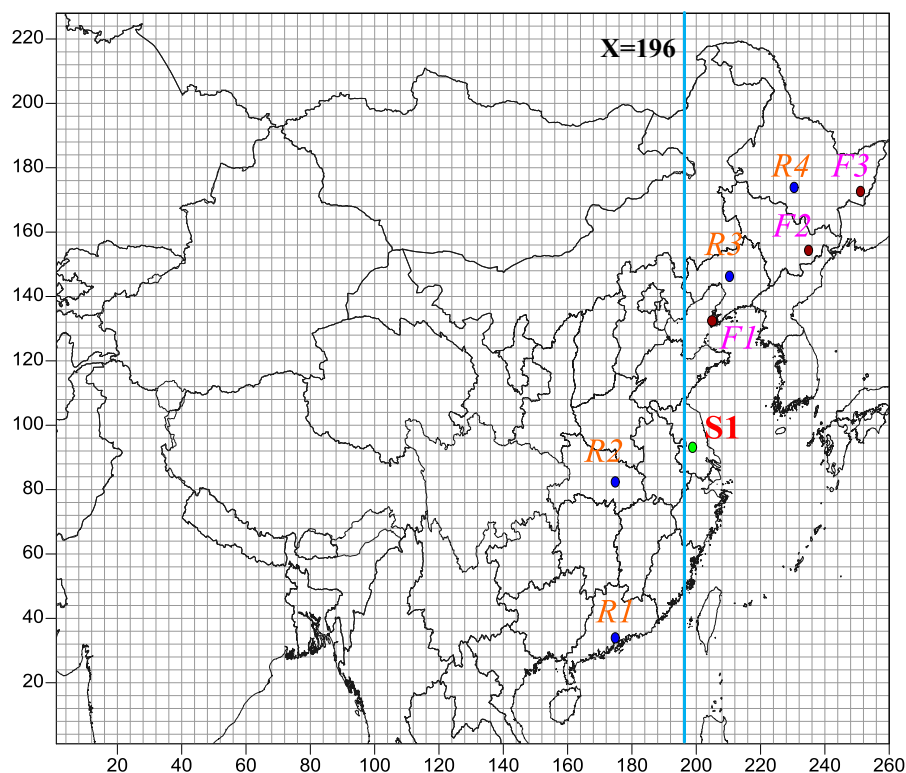
Model experimental design

To better investigate the spatial and temporal characteristics of POPs in the environment, ChnGIPERM was employed to model the environmental behavior and distribution of PCBs from single-source emissions. To our knowledge, this is the first study on the regional distribution characteristics of POPs from single-source emissions in China. The source site (corresponding grid cell S1 (190, 96), Fig. 1) is located in eastern central China (E $118^\circ22'–119^\circ14'$; N $31^\circ14'–32^\circ37'$). Annual ambient air temperature at the source site ranges between -13.1 and 39.7°C with an annual average temperature and rainfall of 15.4°C and 1106 mm, respectively. Figure 1 shows all four receptors in the latitudinal direction, which are from $\sim 22.2^\circ\text{N}$ (R1) to $\sim 46.4^\circ\text{N}$ (R4); the four sites and corresponding grid cells are R1 (174, 37), R2 (180, 81), R3 (214, 149), and R4 (227, 173) from south to north.

In general, PCBs have 209 congeners, and CB28, CB 52, CB 101, CB 118, CB 138, CB 153, and CB 180 are called indicative PCBs because their physicochemical properties have been characterized more thoroughly than those of other PCB congeners. These PCBs can represent different PCB homologues. Hence, we selected representative low molecular weight (LMW) and high molecular weight (HMW) PCB congeners for modeling, including CB28, CB101, CB153, and CB180.

Numerical simulations were successively performed using an assumed emission rate (1 kg d^{-1} from source S1) for a single year only (1965) and can

Fig. 1 The model domain and source (S1, green circle), receptors (R1, R2, R3, and R4, blue circle), and observation sites (F1, F2, and F3, deep red circle). Solid light blue lines were used to extract latitude transect data along the emission source to observation sites



be thought of as a single emission event. Meteorological and environmental data (“[Physicochemical properties of PCB congeners](#)” section) from 1965 through 2010 were used to simulate potential for LRAT and fractionation phenomena. The spatial–temporal distribution of the single emission year (1965) was continually modeled until 2010. The initial concentration of the selected PCB congeners in soil was assumed to be zero.

Physicochemical properties of PCB congeners

The model input data included gridded daily meteorological data (temperature, precipitation, wind direction and speed), soil parameters (density, organic carbon content, and porosity), surface features (urban, dry cropland, paddy field, grass, forest, uncultured land and water types), and physicochemical properties of the chemicals. Relevant physicochemical property data for the four representative congeners CB28, CB101, CB153, and CB180 are listed in Table 1.

Results and discussion

Atmospheric transport of PCBs

Many factors can affect the environmental distribution of POPs, such as meteorological conditions (temperature, wind direction and speed, precipitation, etc.) and environmental parameters (including soil type, soil organic carbon content, vegetation type). Atmospheric aerosols can also influence the presence of POPs in the atmosphere through deposition (dry and wet deposition) of absorbed POPs, which can influence the atmospheric transport of POPs from sources to remote regions.

In a previous study, the characteristic travel distance (CTD) was used to assess the transport potential of chemicals (Bennett et al. 1998). The CTD is the distance from the source region at which the concentration in the Lagrangian cell is reduced to $1/e$ of its initial value, i.e., by 63% (Bennett et al. 1998). The concept of stickiness was introduced by Beyer et al. (2000) and indicates the fraction of a substance that will partition into and remain in a nonmobile medium, usually soil, and thus retard transport. In this

Table 1 The physicochemical properties of the four PCB congeners used in the model (reference temperature 25 °C)

Parameters (unit)	CB28	CB101	CB153	CB180	References
Molar mass (g mol ⁻¹)	257.5	326.4	360.9	395.3	Mackay et al. (2006)
Molar volume: V_M (cm ³ mol ⁻¹)	247.3	289.1	310.0	330.9	Mackay et al. (2006)
Henry's law constant (Pa m ³ mol ⁻¹)	31	27	23	22	Mackay et al. (2006)
Melting point (°C)	57	77	103	110	Mackay et al. (2006)
Aqueous solubility (g m ⁻³)	0.16	0.01	0.001	0.0002	Mackay et al. (2006)
Vapor pressure (Pa)	0.034	0.0033	0.00067	0.00013	Mackay et al. (2006)
Organic carbon–water partition coefficient Log K_{oc} (dimensionless)	4.28	5.99	6.53	6.97	Mackay et al. (2006)
Octanol–water partition coefficient Log K_{ow} (dimensionless)	5.67	6.37	6.88	7.20	Mackay et al. (2006)
Degradation rate in air: K_a (h ⁻¹)	3.50×10^{-7}	4.08×10^{-4}	1.26×10^{-4}	3.50×10^{-8}	Wania et al. (1999)
Degradation rate in water: K_{rw} (h ⁻¹)	4.07×10^{-5}	1.26×10^{-5}	1.26×10^{-5}	1.26×10^{-5}	Mackay, et al. (1992)
Degradation rate in sediment: K_{rsc} (h ⁻¹)	1.26×10^{-5}	1.26×10^{-5}	1.26×10^{-5}	1.26×10^{-5}	Mackay et al. (1992)
Base degradation rate in the six kinds of soil: K_{s0} (h ⁻¹)	$1.26E \times 10^{-5}$	1.26×10^{-5}	1.26×10^{-5}	1.26×10^{-5}	Mackay et al. (1992)
Activation energy of deg. by OH radicals (J mol ⁻¹)	8314	8314	8314	8314	Wania et al. (1999)
Activation energy for freshwater/seawater (J mol ⁻¹)	30000	30,000	30,000	30,000	Beyer et al. (2002)
Activation energy for air (J mol ⁻¹)	13,700	15,000	15,400	17,800	Beyer et al. (2002)
Activation energy for soil (J mol ⁻¹)	30,000	30,000	30,000	30,000	Beyer et al. (2002)

study, we introduce the concept of atmospheric transport time (ATT) to interpret the propensity (on a national scale) of chemicals to travel and to explore the influence of prevailing wind in winter and summer on the source–sink relationships and spatial distributions of volatile organic pollutants. The ATT was calculated using ChnGIPERM with single-source emissions of chemical starting on 1 January (winter-time) and 1 July (summertime), respectively. The ATT is derived as simply the estimated time taken (days) for the emitted pollutant to reach the receptor.

Figure 2a depicts the ATT in winter from the emission source (S1) to each receptor, R1, R2, R3, and R4, which was 20, 21, 46, and 66 days from the south (low latitude) to north (high latitude), respectively. In contrast, the ATT was 46, 22, 17, and 21 days in the

summer from the low latitude (R1) to high latitude (R4), respectively (Fig. 2b). Some studies have reported that the Westerlies and East Asian monsoons are key factors that can influence the transport patterns of volatile organic pollutants in China (Zhang et al. 2011), and our results add further evidence to corroborate this. The Hadley Cell and Ferrel Cell dominate in different seasons; both move northward in the summer and southward in the winter (Zeng et al. 2011). These atmospheric cycles explain why trans-Pacific transport of pollutants from Asia to the western seaboard of North America is always accompanied by strong westerly winds that are most dominant during winter (Bey et al. 2001; Liu et al. 2003; Zhang et al. 2008). Tian et al. (2009) showed that The East Asian summer monsoons provide the main atmospheric pathways and

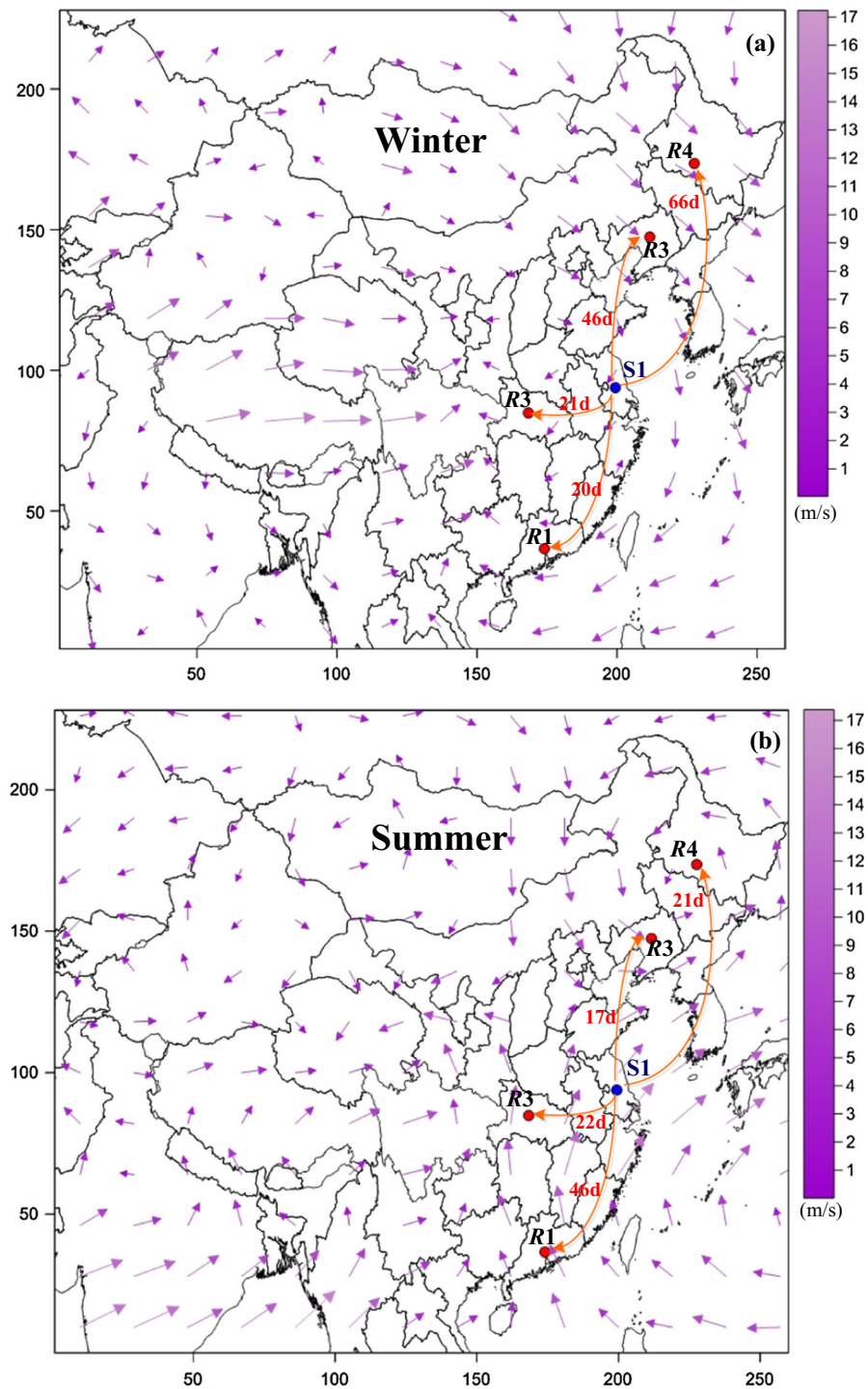


Fig. 2 Atmospheric transport time from the source to receptors in winter (a) and summer (b) for CB28

summer outflows of α -hexachlorocyclohexane (α -HCH) over East Asia.

Gas/particle partitioning of PCBs

Atmospheric transport is the main pathway by which most POPs move from source emissions to more remote regions. The environmental fate of POPs is primarily governed by their partitioning between gases and particles (Lohmann et al. 2000). The gas-particle partitioning of chemicals depends on ambient temperature, the nature of ambient aerosols, and the sorption and desorption behaviors of the chemical and aerosol (Pankow and Bidleman 1991; Pankow 1994). The tendency of chemicals to evaporate into the atmosphere is primarily controlled by their vapor pressure (Mackay 2001). Among the congeners included in this study, the most volatile (CB28) has a vapor pressure 4 orders of magnitude higher than the least volatile congener studied (CB180); $-50\text{ }^{\circ}\text{C}$ *cf.* $50\text{ }^{\circ}\text{C}$ (Falconer and Bidleman 1994). The subcooled liquid vapor pressure has been used to describe both

absorptive and adsorptive partitioning (Pankow and Bidleman 1991). If an aerosol contains organic matter, absorptive partitioning will be the dominant mechanism. Absorptive partitioning has been successfully described using the octanol–air partition coefficient (K_{OA}) and organic matter content (Harner and Bidleman 1998).

Figure 3 shows that the proportion of each congener sorbed to particles increases with increasing latitude, and the amount of chlorine increases. In winter, this relative increase in particle association is seen for all PCB congeners studied (due to the relatively low temperatures). A reverse trend is observed in the summer for all PCB congeners due to the relatively high temperatures. This effect is especially pronounced for the more volatile congeners; for example, the difference in particle-associated CB28 between R1 and R4 was $\sim 20\%$ in December. During the summer, the presence of the more volatile congeners in the atmosphere is dominated by the gas fraction. All these factors are related to temperature, and the gaseous form can facilitate longer distance transport compared to the

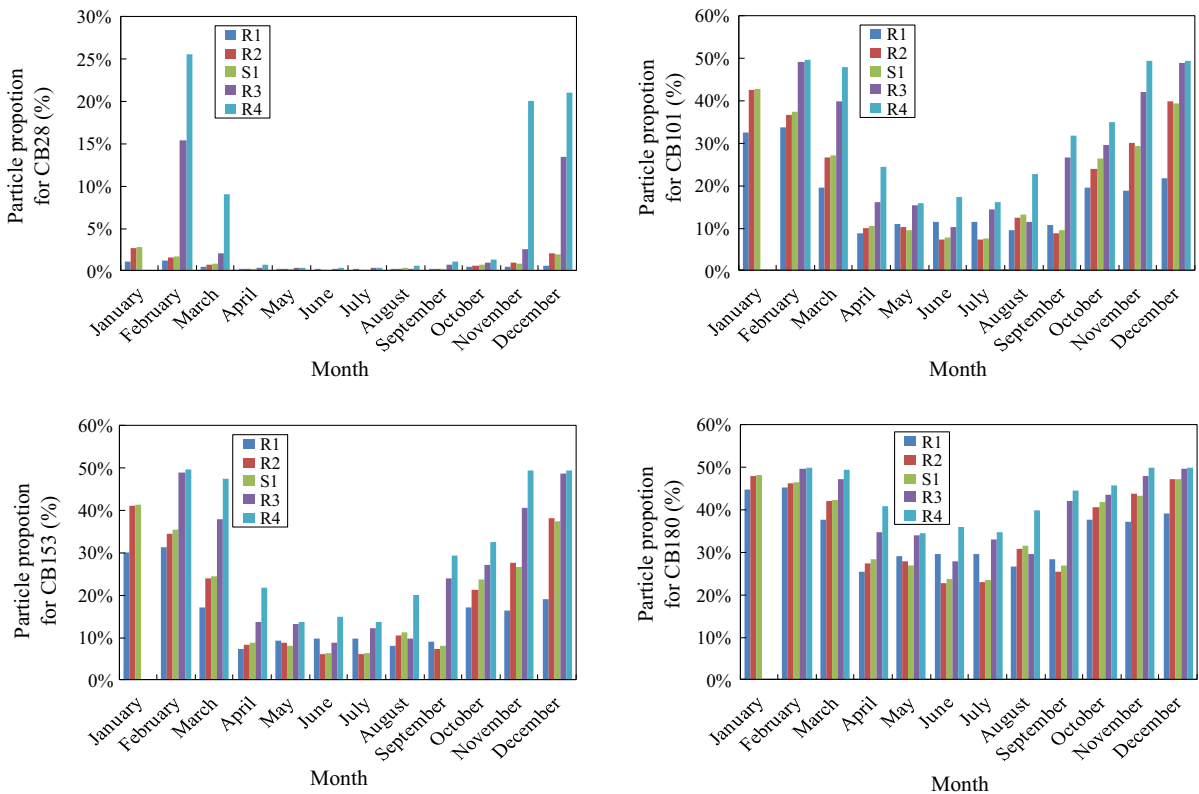


Fig. 3 The particle proportions from low to high latitudes for PCB congeners

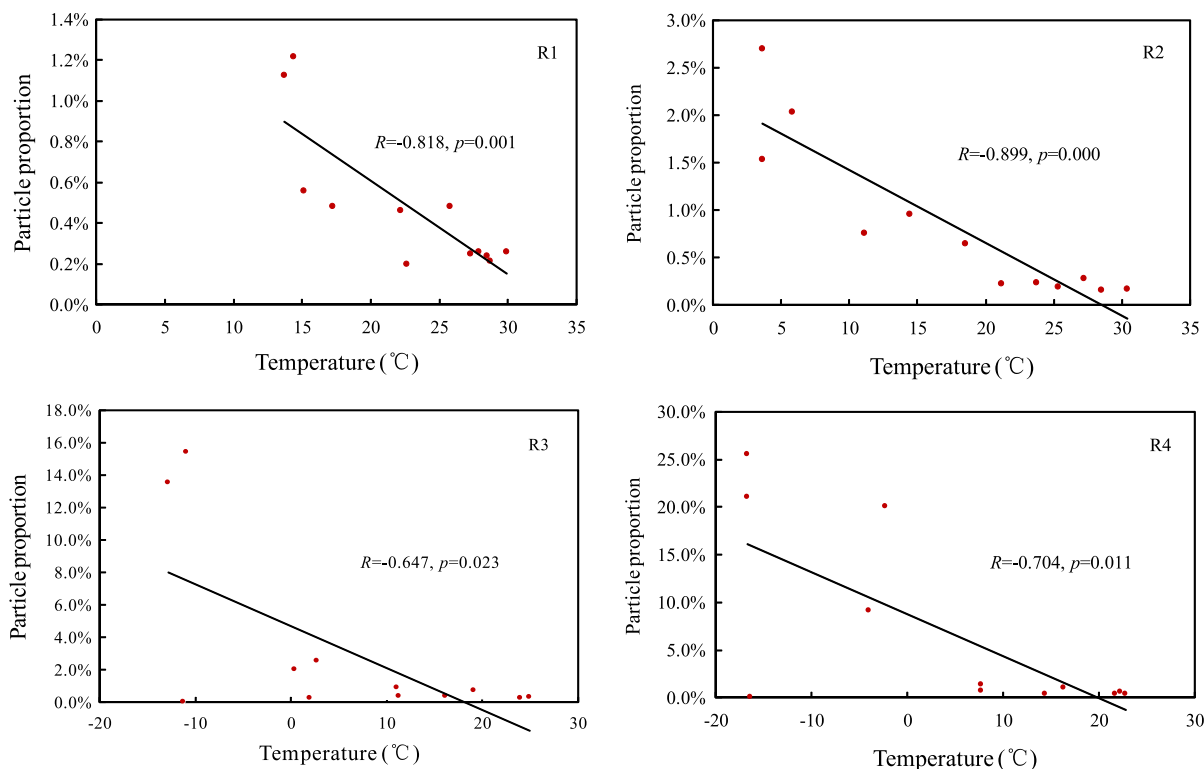


Fig. 4 The relationship between the temperature and particle proportion of CB28 at different receptors

particulate form, which is an important reason for the relatively large differences in the ATT between winter and summer for the farthest receptors. For the lower volatility PCB congeners, the transport capacity is relatively low as their fractionation is more heavily dominated by the particulate form, this effect is particularly pronounced for CB180. A significant negative relationship was elucidated between ambient temperature and the proportion of total atmospheric CB28 present in particulate form at each receptor (Fig. 4).

Primary fractionation effect

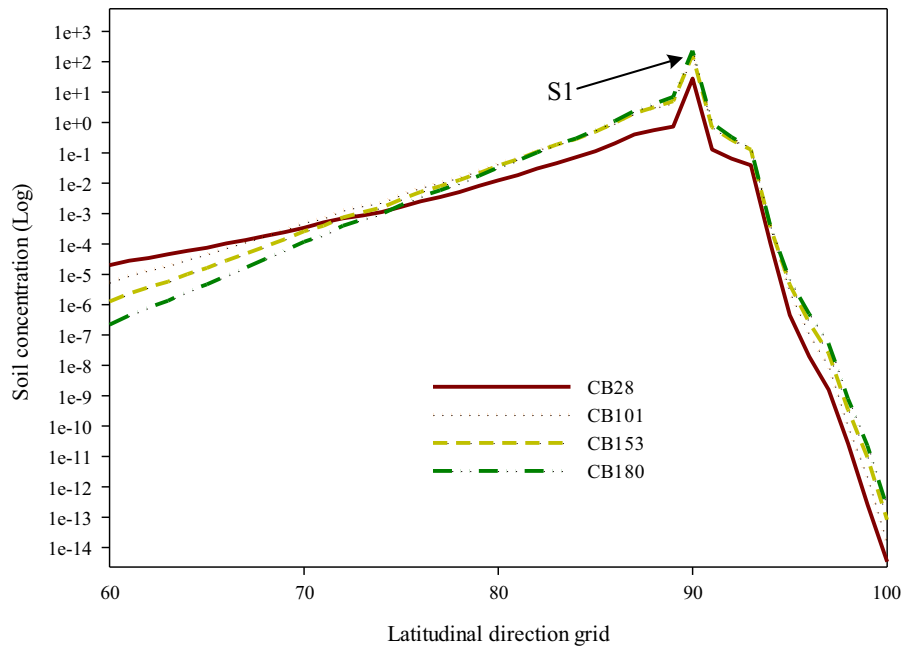
The primary fractionation effect for individual PCB congeners

To better understand primary fractionation, a latitude transecting line (from grid cell 60 to 100) along the emission source was selected to investigate this phenomenon within the ChnGIPERM platform. Figure 5 shows that the highest concentrations for all PCB congeners in soils developed close to the

emission source (S1), with gradients of declining concentration with distance from source. These spatial patterns represent the general characteristics of primary fractionation. Within this general distribution pattern, LMW congeners tended to be transported further compared to HMW congeners simply because LMW congeners are easier to transport and undergo relatively more OH radical reactions in the atmosphere and volatilization (Totten et al. 2002).

The decline in predicted concentrations of each PCB congener in soils north of S1 (grid cells 90 to 100) was steeper than the same estimates for locations south of S1 (grid cells 90 to 60) due to how the January emission period is simulated in ChnGIPERM; the Asia winter monsoon is the prevailing wind. Similarly, concentrations of the HMW congeners tended to be elevated with respect to LMW congeners in soils north of S1, while this pattern was reversed in soils south of S1 at around grid cell 75. This pattern may indicate the existence of other primary fractionation characteristics, especially for a set of congeners, indicating that LMW CB28 can be transported farther than HMW CB180.

Fig. 5 Primary fractionation of individual PCB congeners in January



The primary fractionation effect for a set of PCB congeners

Another significant characteristic of primary fractionation is that the ratio of LMW PCB increases with distance. Therefore, the CB28, CB101, CB153, and CB180 were selected as modeled congeners to reveal the fractionation effect, and it can be expressed as follows:

$$P_i = \frac{C_i}{\sum_{i=1}^n C_i} \tag{1}$$

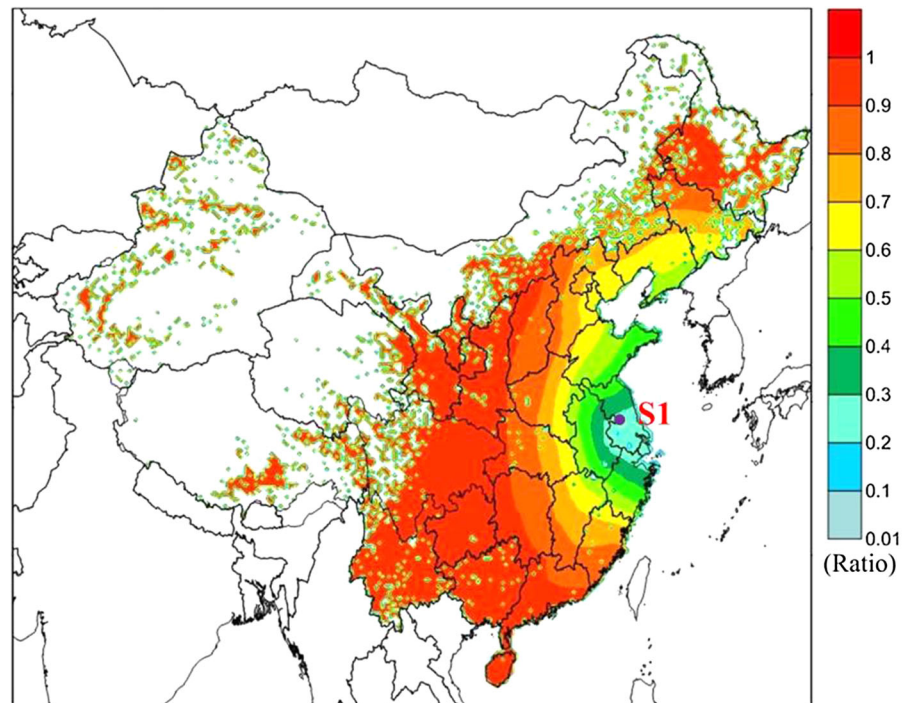
where P_i is the ratio fraction, C_i is the soil concentration of the individual PCB congener (ng g^{-1}), i is the modeled PCB congener, and n is the number of modeled PCB congeners. In this study, the ratio fraction considered was between CB28 and the total modeled PCB congeners.

In general, more volatile compounds can more easily travel farther to remote regions; in contrast, nonvolatile and immobile substances, such as HMW PCBs, tend to remain in their emission region due to rapid deposition (Wania and Mackay 1993). Figure 6 depicts another primary fractionation characteristic: The CB28 proportion is smaller closer to the emission area than other regions, and the CB28 proportion increases with the distance from the emission area.

Secondary fractionation effect

Given that PCBs have been widely banned under the auspices of the Stockholm Convention, secondary emissions are likely to play an important role in their environmental distribution. After entering the environment, POPs often exchange between environmental media such as soil and air as part of a general propensity to move from warmer to colder regions of the globe, a process often referred to as “grasshopping.” Alternatively, this process can be termed “secondary fractionation” (Li et al. 2010), whereby chemical transport is primarily driven by temperature, SOM, and surface–air partition coefficients of various environmental media or climate zones. In general, POPs participate in several cycles of deposition and re-emission before reaching their final destinations (Wania and Mackay 1996). The status of POPs in soil and air is mainly controlled by air–soil exchange (Jones 1994). POPs continuously exchange through recycling between the atmosphere and terrestrial environments over long periods of time before they reach equilibrium (Backe et al. 2004). The flux and direction of the air–soil exchange of POPs vary with time and are influenced by climatic factors, soil properties, and chemical emissions (Backe et al. 2004; Cabrerizo et al. 2011; Wang et al. 2012).

Fig. 6 The fraction of modeled CB28 concentration to the sum of modeled concentrations of all selected congeners in the soil



To explore how ChnGIPERM models secondary fractionation effects, the three grid sites F1 (206,131), F2 (236,151), and F3 (255,178) were selected over the whole model domain (Fig. 1). Within these grid cells, data for CB28 indicated that primary fractionation occurred in the emission period in 1965, and subsequently, an irregular trend was observed among the three sites (Fig. 7). Over 20 years after emissions stopped, the CB28 appeared again at the end of the 1980s with an inverse trend from F1 to F3. This phenomenon indicated that ChnGIPERM is able to model secondary fractionation as long-term accumulation in relatively cold climates, but that secondary fractionation is a relatively slow process for CB28 or other relatively volatile POPs. In reality, secondary fractionation can be thought of as a continuous process of multiple deposition/volatilization steps before a chemical reaches its final destination.

To further explore evidence of secondary fractionation within the ChnGIPERM model environment, the SOC, annual average temperature, and soil density were extracted from the environmental and soil property data of the model (Table 2). Soil density is almost the same at the three observation sites (F1, F2, F3), and thus this parameter has limited influence the

concentrations of CB28 at the three sites. However, sites F1 and F2 have a relatively large difference in temperature, but only a 1% organic carbon content difference between the sites. The vapor pressures of many POPs increase three- to fourfold for each 10 °C increase in temperature (Spencer and Cliath 1983), and thus, the volatilization of CB28 has stronger at F1 than F2 and F3. With the same annual average temperature, F3 has a stronger soil absorption capacity for CB28 than F2 because the SOC content can strongly affect its environmental distribution and re-emission from soil. Thus, within ChnGIPERM, both temperature and SOC content play a crucial role in the secondary fractionation of volatile organic pollutants over long-term evolution. However, the secondary fractionation related to soil–air exchange of volatile organic pollutants which is mainly subjected to the temperature and SOC. Cousins et al. (1999) also identified that the temperature and SOC can influence the soil–air exchange of chemicals through calculated soil–air fugacity quotients for selected semivolatile organic compounds (SVOC) in the UK.

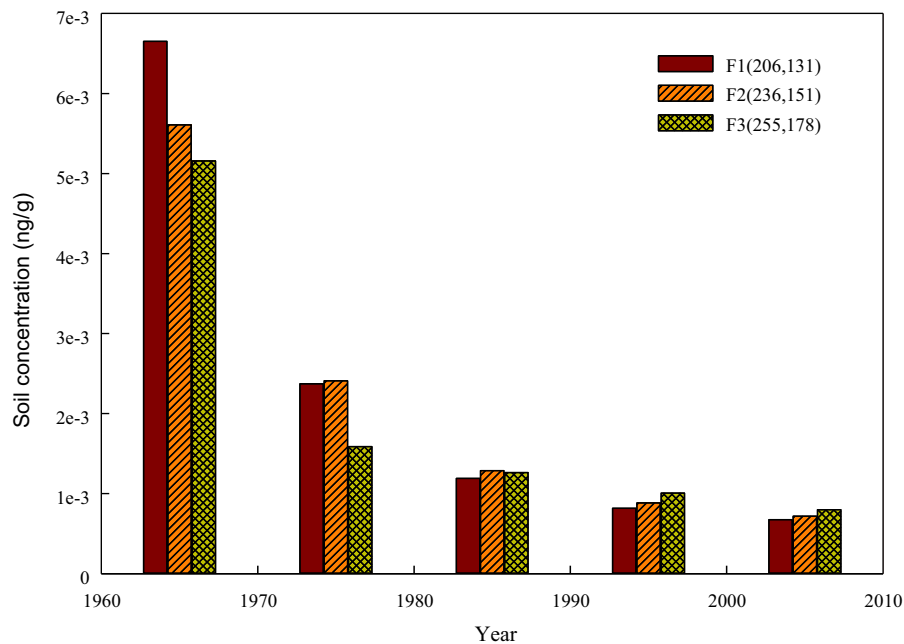


Fig. 7 Variations of CB 28 concentrations in soil over time: evidence of secondary fractionation effect

Table 2 Annual average temperatures and soil properties of the modeled observation sites

	F1 (206,131)	F2 (236,151)	F3 (255,178)
Temperature (K)	284	274	275
SOC content (%)	2.55	1.56	8.89
Soil density (kg m ⁻³)	1345.2	1373.9	1353.5

Conclusion

Using ChnGIPERM with single-source emissions, a modeling investigation was carried out to explore the environmental behavior and distribution of PCB congeners modeled over a 46-year time series. This study showed that atmospheric transport is influenced by meteorological factors, such as Westerlies and East Asia monsoons, and introduced the concept of ATT, which appears promising for describing this phenomenon. The partitioning of PCB congeners between gas and particle forms was mainly controlled by temperature, which can further affect the ATT. In this study, we confirmed that ChnGIPERM adequately models both primary and secondary fractionation effect from single-source PCB emissions, with results in line with our mechanistic understanding.

ChnGIPERM represents secondary fractionation processes of PCBs as slow multiple steps in the environment. This has implications for environmental management and remediation of POPs, given the significant legacy of even those substances which have been banned for many decades.

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Compliance with ethical standards

Conflict of interest The authors declare that they have no competing interests.

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