



Nutrient distribution and structure affect the acidification of eutrophic ocean margins: A case study in southwestern coast of the Laizhou Bay, China



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ABSTRACT

The effects of nutrient distribution and structure on the acidity of coastal waters were analyzed based on the data of 48 surface water samples collected in the southwestern coast of the Laizhou Bay and its adjacent rivers (SWLZB) which are heavily influenced by nutrient-laden discharges. The concentration and structure of nutrients varied considerably along the coast owing to different contributors. The studied inshore waters exhibited a sign of acidification. The pH was significantly negatively correlated with the concentration of NO₃-N, NO₂-N, NH₄-N and DSi, but showed no obvious correlation with the concentration of PO₄-P and the ratio of TDN/TDP, DSi/DIN and DSi/PO₄-P, respectively. The results indicated that the distribution of nutrients might well be an important environmental factor affecting the acidification of the SWLZB in warmer months.

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1. Introduction

It is well-known that nutrient over-enrichment is a major environmental problem in many coastal areas around the world (Cloern, 2001; Seitzinger et al., 2005; Song, 2011). Coastal areas are regions with high population density and intense human activities, accounting for ~30% of the total net oceanic primary production (Alongi, 1998; Gattuso et al., 1998; Dürr et al., 2011; Song, 2011). Large amounts of nutrients have been transported from land to coastal areas by rivers in the past decades due to the rapid increase of human activities, resulting in environmental deterioration and biogeochemical process variation (Seitzinger et al., 2005; Halpern et al., 2008; Qu and Kroeze, 2010). The concentration and composition of nutrients could not only affect the biomass and composition of phytoplankton but also cause large-scale algal bloom when there is too much of them (Gao and Song, 2005; Song, 2011; Smetacek and Zingone, 2013; Xing et al., 2015). Benefiting from the nutrient enrichment, many microbial species are flourishing in eutrophic coastal seas, and their respiration could release much CO₂ which may lower the seawater pH in these areas (Cai et al., 2011; Wallace et al., 2014). In addition, some nutrients may exist in the form of acid or weak base, for instance, N may exist as HNO₃ or NH₃ in some particular situations, which may affect the acidity of seawaters directly. Ocean acidification (OA) induced by human activities, especially in coastal regions, has aroused great concern in recent years

(Caldeira and Wickett, 2003; Canadell et al., 2007; Quéré et al., 2013). From the preindustrial period to the 1990s, the mean pH of global surface oceans had decreased from approximately 8.2 to 8.1, corresponding to a 26% increase in the acidity of the oceans as measured by the concentration of hydrogen ion (Gattuso and Hansson, 2011). With few exceptions, this change exceeded that in any period in the last 300 million years, and is projected to continue to influence the world oceans of greater depths progressively (Caldeira and Wickett, 2003). The majorities of marine bios reside in the surface ocean and are exposed to this progressive increase of seawater acidity. Microbes collectively support marine food webs, and the activities of biota influence the composition of dissolved chemicals and particulate organic materials and moderate the exchange of nutrient elements such as carbon, nitrogen, phosphorus between the ocean waters (Cooley et al., 2009), and all these processes are subject to OA. Although OA and nutrients in coastal waters have intimate interactions, at present there is little research concerning the effects of nutrient distribution and structure on OA.

The effects of nutrient distribution and structure on OA may be evidenced in waters of the southwestern coast of the Laizhou Bay (SWLZB). There are ten rivers (Yihonghe River, Guanglihe River, Zhimaihe River, Xiaoqinghe River, Mihe River, Yuhe River, Bailanghe River, Dihe River, Weihe River, and Jiaolaihe River) flowing into the Laizhou Bay from its southwestern coast, most of which are small and seasonal. Due to abundant seawater resources and underground brine resources, a chemical industrial base called the Weifang Binhai Economic Development Zone is located along its southwestern coast. Over 400 chemical enterprises are located nearby and >150 kinds of chemical products are

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manufactured in this area. Thus, large quantities of CO₂, SO₂ and NO₂ were produced in these chemical industrial activities (Qin, 2010), which may affect the acidity of the coastal waters of the SWLZB. In addition, the river waters carry large amounts of nutrients derived from many point sources such as industrial effluents and agricultural runoff, non-purified or insufficiently purified wastewaters discharged into the inshore waters of this area (SOA, 2009; SOA, 2012). Therefore, the aquatic environment of the SWLZB is in a eutrophic state (Zhang et al., 2015). Different rivers have different nutrient concentration and composition which may change the concentration and structure of nutrients in the inshore waters along the coast. Such a condition in the SWLZB provided the opportunities for the analysis of the interactions between nutrients and OA.

This study tried to determine the concentrations of total dissolved nitrogen (TDN), dissolved inorganic nitrogen (DIN), nitrate nitrogen (NO₃-N), nitrite nitrogen (NO₂-N), ammonium nitrogen (NH₄-N), dissolved organic nitrogen (DON), total dissolved phosphorus (TDP), reactive phosphorus (orthophosphate-P; PO₄-P), dissolved organic phosphorus (DOP), and dissolved silicate (DSi) in the surface waters of the SWLZB and the pH in the same sites with a purpose of attempting to analyze the interactions between the distribution or structure of nutrients and OA.

2. Materials and methods

2.1. Sample collection

The research was carried out during 23 September to 1 October 2012, which is after the peak period of the rainy season. The sampling stations were arranged in the flowing direction of the major rivers in this area covering about 15–20 km from the high tide mark to the land and about 10 km from the high tide mark to the sea, and keeping about 2–3 km from each other (Fig. 1). The river stations arranged in ten rivers of the SWLZB coast, which are Yihonghe River (YHH), Guanglihe River (GLH), Zhimaihe River (ZMH), Xiaoqinghe River (XQH), Mihe River (MH), Yuhe River (YH), Bailanghe River (BLH), Dihe River (DH), Weihe River (WH), and Jiaolaihe River (JLH). The sampling sites in the inshore waters formed five transects, here named as L-transect (sites L1, L2, L3, L4 and L5), K-transect (sites K1, K2 and K3), J-transect (sites J1, J2, J3 and J4), I-transect (sites I1, I2 and I3) and H-transect (sites H1, H2 and H3). A total of 48 water samples were collected from the surface (depth of ~0.5 m) using well cleaned polyethylene bottles (pre-conditioned for 24 h in 1:10 nitrate acid solution and

rinsed well with 18.2 MΩ-cm deionized water). The samples were stored in a cooler box with ice bags, and then refrigerated at 4 °C within 12 h until further analysis. All the samples were filtered with cellulose acetate membrane filters (Whatman®, 0.45 μm) before nutrient determination.

2.2. Analytical methods

TDN, NO₃-N, NO₂-N, NH₄-N, TDP, PO₄-P and DSi were measured using continuous flow analyzer (SEAL AutoAnalyzer 3, German). The analytical precision for TDN, NO₃-N, NO₂-N, NH₄-N, TDP, PO₄-P and DSi was assessed by multiple analysis of standards and replicates. The precision of the analyses of standard solution was better than 5%, which was consistent with the previous report of Liu et al. (2011); all analyses of the samples were carried out in duplicate and the relative standard deviations were <10%, and the results were expressed as the means. The concentrations of dissolved inorganic nitrogen (DIN) were the sum of NO₃-N, NO₂-N and NH₄-N. The concentrations of dissolved organic nitrogen (DON) and phosphorus (DOP) were estimated by the subtractions of DIN from TDN and PO₄-P from TDP, respectively.

The pH values were determined in situ using a portable Orion Star-A combined instrument equipped with an Orion® 8107UWMMMD Ross combination electrode (Thermo Scientific, Singapore) for the precision of pH analysis. The pH buffers adopted were three NIST (National Institute of Standards and Technology, the US Department of Commerce)-traceable pH buffers. According to the results of Zhai et al. (2012), in the Bohai Sea, the pH values on the total-hydrogen-ion scale (pH_T) were lower than the NIST-traceable pH values by 0.143 ± 0.006 pH units (mean ± standard deviation, $n = 73$). Based on this result, we transferred the NIST-traceable pH data to the total-hydrogen-ion scale (pH_T).

2.3. Statistical analysis

All data were tested for normality with the Shapiro–Wilk test before analysis. The Spearman correlation-coefficient was used to analyze the relationship between pH_T and nutrient concentrations or ratios. In this research, a value of $P < 0.05$ was considered to indicate a significant difference in all statistical analysis. Multiple regression analysis was used to estimate the coefficients of the linear equation, based on which the value of the dependent variable could be well predicted. The analyses were performed through the Origin statistical package version 7.0.

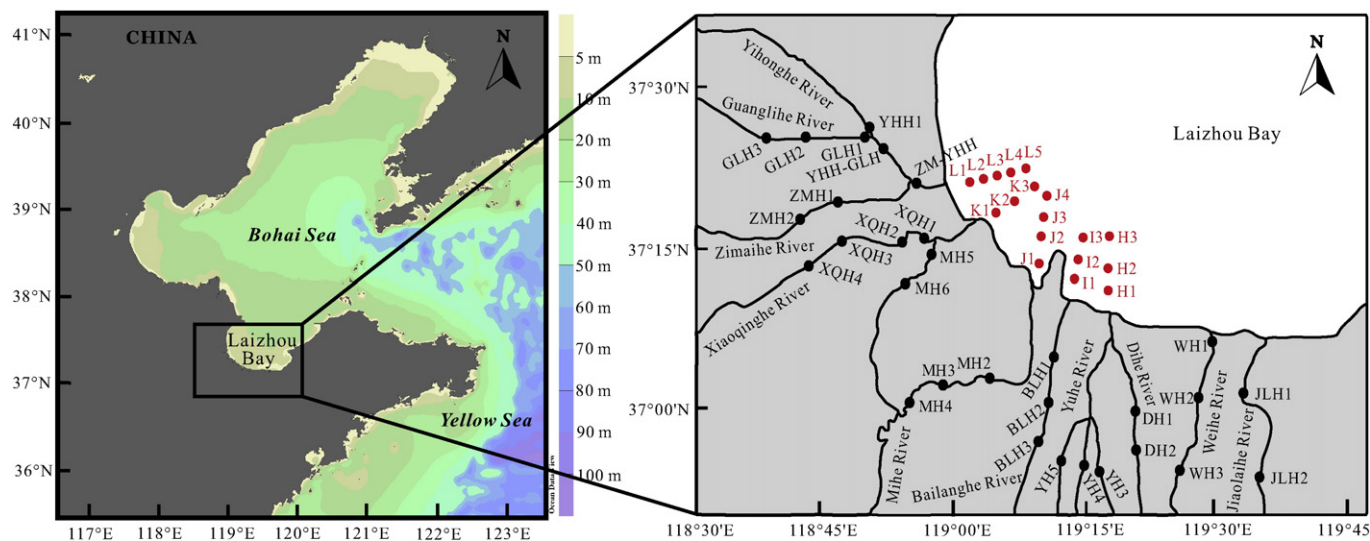


Fig. 1. Location of sampling stations in the coastal waters of the SWLZB. The marine sampling stations were shown in red dots and the riverine ones were shown in black.

3. Results and discussion

3.1. The acidity (pH) in the studied region

The pH values on total-hydrogen-ion scale (pH_T) in the studied areas varied from 7.16 to 8.64 with an average of 7.81 (Fig. 2). It could be seen in Fig. 2 that the pH_T values in 81.5% of the total sampling sites were below the average pH_T values of the world oceans (varied from 7.9 to 8.3 with an average of 8.1 on the total-hydrogen-ion scale) (Ryuichi et al., 2013). The pH_T values in the inshore waters varied from 7.56 to 8.26 with an average of 7.93 (Fig. 2), which increased in the seaward direction (Fig. 2). It could be seen in Fig. 2 that the pH_T values in 88.9% of the marine sites were below the average pH_T values of the world oceans (8.1) (Ryuichi et al., 2013). Moreover, the decadal changes of pH_T values in the Laizhou Bay surface waters demonstrated a decreasing trend, although these data were obtained in different areas and months in the Laizhou Bay, from 8.11 in 1987 to 7.93 in 2012 (Fig. 3), indicating a sign of gradual acidification (Figs. 2 and 3). The different inshore transects exhibited different degrees of acidification: the pH_T values of L-transect, K-transect, J-transect, I-transect and H-transect were in the range of 7.69–8.00, 7.91–8.06, 7.84–8.03, 7.56–8.18 and 7.96–8.26, with a mean value of 7.87, 7.98, 7.96, 7.78 and 8.10, respectively, and all of the five marine transects were acidified to some extent compared with those reported values that were shown in Fig. 3. Among them, the I-transect was the most seriously acidified one, followed by L-transect, J-transect, K-transect and H-transect.

3.2. Distribution characteristics of nutrient species

Spatial distributions of nutrient species including TDN, DIN (NO_3-N , NO_2-N and NH_4-N), DON, TDP, PO_4-P , DOP and DSI in the inshore transects are shown in Fig. 4. Although the distribution of different nutrient species in the whole inshore area showed different patterns, a common distribution characteristic was identified that the order of their corresponding concentrations was K-transect > H-transect > L-transect > J-transect > I-transect for most of the studied inorganic nutrient species. Such a pattern was generally corresponding to the nutrient concentrations in the adjacent rivers (Fig. 4), indicating that the concentrations of nutrients in the SWLZB could be intimately associated with river inputs. In order to demonstrate the relationship between nutrient behavior and river inputs in the studied areas, the average concentrations of nutrient species in the SWLZB adjacent rivers were calculated and shown in Fig. 4. It could be seen that TDN concentrations in the western rivers were generally higher than those in the eastern rivers (except for the YH), and the DIN concentrations were lower than DON concentrations except for the XQH. The highest DIN concentration was found in the XQH (Fig. 4a). In the L-transect adjacent rivers (LTAR), the concentrations of NH_4-N were higher than those of NO_3-N and NO_2-N ; while in other four rivers, NO_3-N was the species with highest

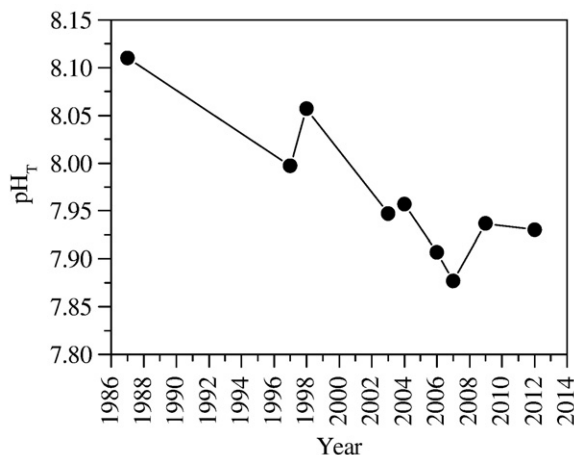


Fig. 3. Decadal change of average pH_T in the Laizhou Bay. The pH_{NIST} values from literature were converted to pH_T according to the pH_{NIST} and pH_T relationship in Zhai et al. (2012). Data sources: 1987: Jiang et al., 1991; 1997: Mi et al., 2001; 1998: Cheng, 2004; 2003: SOA, 2004; 2004: SOA, 2005; 2006: Ying et al., 2014; 2009: SOA, 2010; 2012: present study.

concentration (Fig. 4b). Specially, the YH has the highest TDN and DON concentration, and DON concentrations were much higher than DIN concentrations in this river (Fig. 4b), such a special situation probably associated with the fact that the Yuhe River flows through the Weifang Binhai Economic Development Zone which has been suffering from point source pollution of industrial wastes. The higher DON and TDP concentrations in the LAR may be due to the crude oil exploitation activities in the basin of the Yihonghe River, Guanglihe River and Zhimaihe River, which flow through the second largest oilfield (Shengli Oilfield) of China. The higher nutrient concentration in the XQH may be explained by the fact that the Xiaoqinghe River, the largest river in the studied areas, which is originated from Jinan City, flows through eight counties of Shandong Province, with extensive reconstruction for flood control, sailing, irrigation and sewage discharge (Ma et al., 2004), and is heavily influenced by human activities. It was also found that TDP concentrations in the western rivers were generally higher than those in the eastern ones (Fig. 4c), especially for PO_4-P (Fig. 4c). The highest TDP and PO_4-P concentrations were found in the ZMH, which was probably associated with the fact that the Zhimaihe River flows through the rural areas of Guangrao City—an agriculture dominant region where large amounts of phosphate fertilizer are used (Dongying Statistical Yearbook, 2013). In fact, the agriculture of the western part of the studied areas has been generally more developed than that of the eastern part (Dongying Statistical Yearbook, 2013), suggesting that nutrients in this part may be more associated with agricultural activities. Therefore, the transportation and behavior of nutrients in the SWLZB

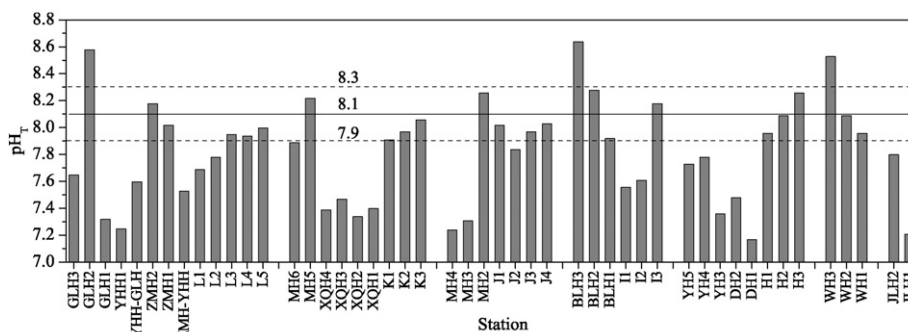


Fig. 2. pH_T distribution in the inshore surface waters of the SWLZB and its adjacent rivers. The pH_T range of the world surface oceans keep in between the two dot lines, and the solid line indicates the present average pH_T of the world surface ocean.

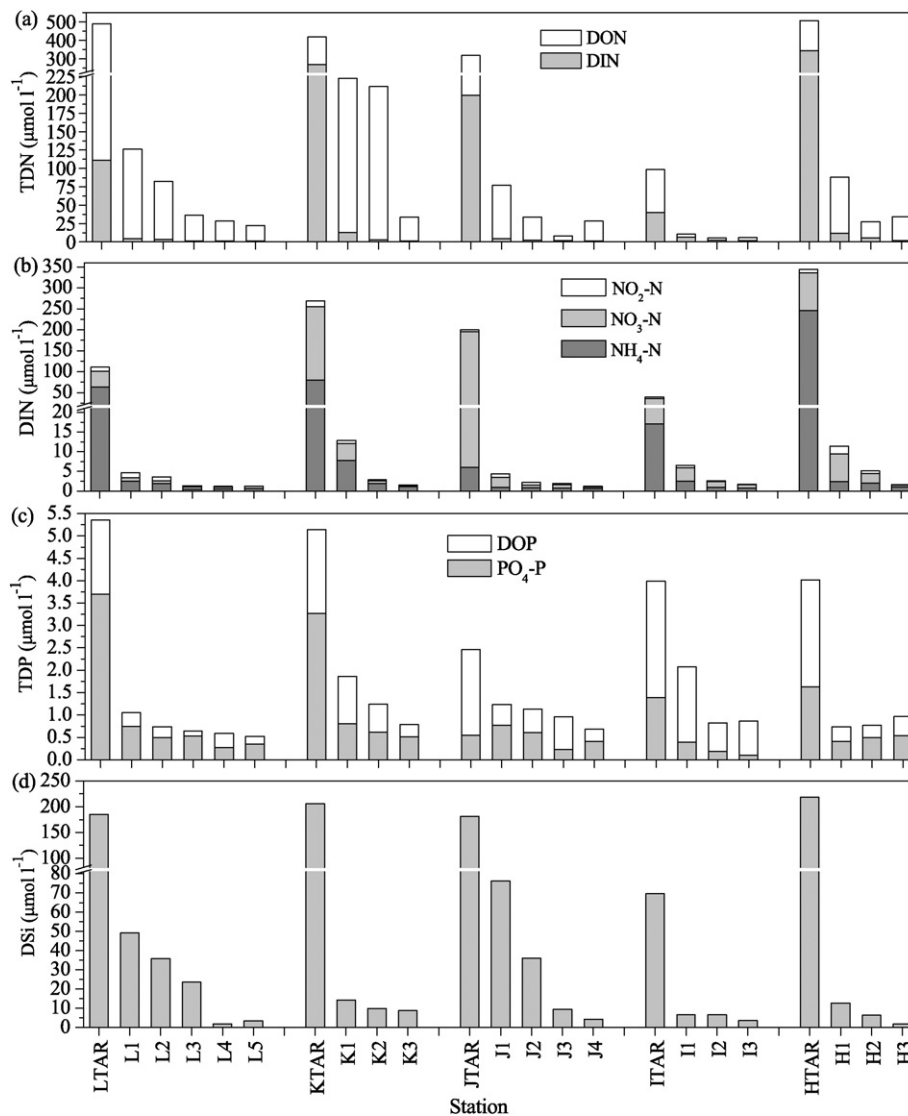


Fig. 4. Distributions of dissolved nitrogen/phosphorus/silicate concentrations and their speciation in the inshore surface waters of the SWLZB and its adjacent rivers. LTAR, i.e. L-transect adjacent rivers, indicates the parameters' mean values of the riverine sites in YHH, GLH and ZMH; KTAR, i.e. K-transect adjacent rivers, indicates the parameters' mean values of the riverine sites MH5, MH6 and those in XQH; JTAR, i.e. J-transect adjacent rivers, indicates the parameters' mean values of the riverine sites in MH except MH5 and MH6; ITAR, i.e. I-transect adjacent rivers, indicates the parameters' mean values of the riverine sites in BLH; HTAR, i.e. H-transect adjacent rivers, indicates the parameters' mean values of the riverine sites in YH and DH.

adjacent rivers were influenced by crude oil exploitation activities, industrial and agricultural activities, similar to that of the Baltic Sea (Kiedrzyńska et al., 2014).

3.2.1. Dissolved nitrogen

The concentrations of dissolved nitrogen constituents in the SWLZB varied greatly along the coast (Fig. 4a). The concentrations of TDN, DIN, $\text{NO}_3\text{-N}$, $\text{NO}_2\text{-N}$, $\text{NH}_4\text{-N}$ and DON in the surface waters of the SWLZB were in the range of 5.29–506.5, 1.17–344.3, 0.31–189.2, 0.05–14.05, 0.43–245.9 and 2.63–378.5 $\mu\text{mol l}^{-1}$, with an average of 126.7, 44.9, 23.5, 2.18, 19.2 and 81.8 $\mu\text{mol l}^{-1}$, respectively. In the studied inshore waters, the concentrations of dissolved nitrogen constituents also varied greatly along the coast (Fig. 4a), which decreased significantly in the seaward direction; K-transect had the highest TDN concentrations, followed by H-transect, L-transect, J-transect and I-transect; L-transect and K-transect were generally enriched with DON relative to DIN, while the reverse was true for the rest of the studied transects. All the sampling transects had the marine sampling sites with DIN concentrations exceeding the upper limit of seawater standard Grade IV

of China for this parameter (2.5 $\mu\text{mol l}^{-1}$). The highest average DIN concentration 5.75 $\mu\text{mol l}^{-1}$ was found in K-transect; the lowest value 2.41 $\mu\text{mol l}^{-1}$, which still exceeded the upper limit of seawater standard Grade I of China for this parameter (1.5 $\mu\text{mol l}^{-1}$), was found in J-transect. So it could be concluded that the SWLZB was enriched with nitrogen.

K-transect had the highest nitrogen concentrations which indicated that nitrogen in this transect was heavily influenced by human activities, because the Xiaoqinghe River is a relatively big one among the rivers discharging into the SWLZB with extensive anthropogenic disturbance as aforementioned (Ma et al., 2004; SOA, 2009; SOA, 2012). This can be illustrated by Fig. 4 in which it could be seen that XQH had the highest nitrogen concentrations among the studied rivers.

The general DIN structure characteristics in L-transect and K-transect were $\text{NH}_4\text{-N} > \text{NO}_3\text{-N} > \text{NO}_2\text{-N}$ (except at L1, L2, L5); in J-transect, I-transect and H-transect, the characteristics were $\text{NO}_3\text{-N} > \text{NH}_4\text{-N} > \text{NO}_2\text{-N}$ (except at J2, J4, H3) (Fig. 4b); specially, in K-transect, DIN was dominated by $\text{NH}_4\text{-N}$, which accounted for >50% of DIN in all the sampling sites (Fig. 4b). These patterns in the studied marine transects

were generally similar to those in their corresponding rivers (Fig. 4b), further indicating that the river inputs could be the dominant source of nitrogen in the studied inshore waters. The concentrations of $\text{NO}_2\text{-N}$ were higher than $\text{NO}_3\text{-N}$ at L1 and L2 which may be due to the point pollution of industrial wastes from Dongying City. K-transect had the highest $\text{NH}_4\text{-N}$ concentrations (mean $3.61 \mu\text{mol l}^{-1}$), which further illustrated the complexity of nitrogen sources in K-transect. The average concentrations of $\text{NO}_2\text{-N}$ in the five marine sampling transects remained at the same level, indicating that $\text{NO}_2\text{-N}$ in the studied inshore waters, on the whole, was very likely to be from non-point sources.

3.2.2. Dissolved phosphorus

The concentrations of TDP, $\text{PO}_4\text{-P}$ and DOP in the surface waters of the SWLZB were in the range of 0.22–6.30, 0.10–3.70 and $0.12\text{--}2.60 \mu\text{mol l}^{-1}$, with an average of 1.68, 0.83 and $0.85 \mu\text{mol l}^{-1}$, respectively (Fig. 4c). The concentrations of dissolved phosphorus decreased in the seaward direction except in H-transect. Except I-transect, all the other transects had a few sampling sites with $\text{PO}_4\text{-P}$ concentrations exceeding the seawater standard Grade I of China for reactive phosphorus ($0.5 \mu\text{mol l}^{-1}$). Although the relatively higher TDN concentration was found in I-transect (I1), the average $\text{PO}_4\text{-P}$ concentration in I-transect was the lowest among the five transects ($0.23 \mu\text{mol l}^{-1}$). Only in K-transect the average $\text{PO}_4\text{-P}$ concentration exceeded the seawater standard Grade I of China for reactive phosphorus ($0.5 \mu\text{mol l}^{-1}$) among the sampling transects. L-transect, K-transect, J-transect and H-transect were generally enriched with $\text{PO}_4\text{-P}$ relative to DOP, while the I-transect were generally enriched with DOP relative to $\text{PO}_4\text{-P}$ (Fig. 4c). Like dissolved nitrogen, these patterns in the studied marine transects were generally similar to those in their corresponding rivers (Fig. 4c), indicating phosphorus in the SWLZB could be originated mainly from river inputs too. As above mentioned, phosphorus in the studied area associated with agricultural activities, which suggested that land use pattern could influence the distribution and speciation of dissolved phosphorus in the SWLZB.

The average concentration of reactive phosphorus in the five sampling transects was relatively low, and such results were similar to the results of the survey conducted in the inshore waters in the Yellow River Estuary by Liu et al. (2012), indicating nutrients in the inshore areas of the Laizhou Bay had high nitrogen concentration and low phosphorus concentration. The K-transect had the highest $\text{PO}_4\text{-P}$ concentration which further indicated that nutrients in that area were heavily influenced by human activities.

3.2.3. Dissolved silicate

The concentrations of DSi in the surface waters of the SWLZB were in the range of $1.82\text{--}218.3 \mu\text{mol l}^{-1}$ with an average of $50.8 \mu\text{mol l}^{-1}$ (Fig. 4d). The dissolved silicate in the inshore waters of the SWLZB varied greatly along the coast, and decreased significantly in the seaward direction. J-transect had the highest DSi concentrations, followed by L-transect, K-transect, H-transect and I-transect (Fig. 4d). The distribution pattern of DSi in the inshore waters was generally similar with that in the corresponding adjacent rivers (see Fig. 4d) indicating DSi in the SWLZB was also influenced by river inputs. The DSi concentrations in most inshore sites remained a mean value of $5.7 \mu\text{mol l}^{-1}$, but its values in some inshore sites from L-transect (L1, L2 and L3) and from J-transect (J1 and J2) were relatively high (mean $44.1 \mu\text{mol l}^{-1}$). The DSi distribution exhibited similar patterns with TDP in the studied rivers (see Fig. 4d). In certain instances silicate minerals originated from crust and their behaviors are less influenced by human activities. However, the crude oil exploitation activities and the sea salt production activities in the bank of the SWLZB could release more silicate minerals from crust into the water system, because of such activities digging more clay and rock (the major ingredients of them are silicate minerals) from underground to the surface.

3.3. Structure of nutrients in the SWLZB

The molar ratios of TDN/TDP, $\text{DIN}/\text{PO}_4\text{-P}$, DSi/DIN and $\text{DSi}/\text{PO}_4\text{-P}$ in the whole studied areas were in the range of 5.1–248.8, 2.5–497.6, 0.5–17.7 and 3.4–932.9 with an average value of 95.4, 86.6, 3.1 and 106.9, respectively.

Nutrient stoichiometry in seawaters is widely used in assessing the nutrient limitation of phytoplankton growth in marine ecosystems. Therefore, nutrient ratios in the inshore waters were specially analyzed in this study. The TDN/TDP ratios of L-transect, K-transect, J-transect, I-transect and H-transect were in the range of 41.7–122.4, 43.8–171.5, 23.2–62.6, 31.7–40.6 and 35.6–168.6 (Fig. 5a) with a mean value of 76.7, 111.6, 39.5, 37.1 and 92.3, respectively. Its spatial variation was very distinct, indicating that different rivers input nutrients of different structures. The J-transect and I-transect had relatively lower TDN/TDP ratios related to higher TDP concentrations. Generally, the studied region had very high TDN concentrations and low TDP concentrations.

The $\text{DIN}/\text{PO}_4\text{-P}$ ratios of the 18 marine sampling stations in the inshore waters ranged from 25.0 to 285.7 (Fig. 5b). Its spatial variation was also distinct. The $\text{DIN}/\text{PO}_4\text{-P}$ ratios of L-transect, K-transect, J-transect, I-transect and H-transect were in the range of 25.0–72.1, 30.6–160.9, 29.3–82.3, 132.9–177.1 and 31.2–285.7 (Fig. 5b) with a mean value of 47.8, 79.6, 51.3, 157.7 and 139.8, respectively. The $\text{DIN}/\text{PO}_4\text{-P}$ ratios for all the sampling stations were higher than 22:1, which means a P limitation condition according to the research of Justić et al. (1995). The highest ratio (285.7) was found in H-transect (Fig. 5b). Except for I-transect, the spatial distribution of $\text{DIN}/\text{PO}_4\text{-P}$ ratio for the other transects were similar to that of the TDN/TDP ratio.

DSi/DIN ratios in the inshore waters mainly ranged from 0.1 to 0.5 in K-transect, I-transect and H-transect, and a relatively wide range of 0.5 to 1.7 was found in L-transect and J-transect (Fig. 5c). The DSi/DIN ratios of about 60% of the marine sampling sites were <0.5 and the low ratios were mainly found in I-transect and H-transect. The DSi/DIN ratios of L-transect, K-transect, J-transect, I-transect and H-transect were in the range of 0.2–1.1, 0.1–0.6, 0.4–1.7, 0.1–0.3 and 0.1–0.2 (Fig. 5c) with a mean value of 0.9, 0.3, 1.1, 0.2 and 0.1, respectively.

Ratios of $\text{DSi}/\text{PO}_4\text{-P}$ in the inshore waters ranged from 3.5 to 99.5 with three of the five sampling transects being higher than 22:1 (Fig. 5d). The high ratios were mainly found in L-transect and J-transect. Different from the wide variation of TDN/TDP and $\text{DIN}/\text{PO}_4\text{-P}$ in K-transect, there were no significant variations of $\text{DSi}/\text{PO}_4\text{-P}$ in K-transect. The $\text{DSi}/\text{PO}_4\text{-P}$ ratios of L-transect, K-transect, J-transect, I-transect and H-transect were in the range of 7.2–66.9, 16.4–17.8, 10.8–99.3, 16.3–36.1 and 3.5–31.9 (Fig. 5d) with a mean value of 39.8, 17.3, 52.6, 28.3, and 16.0, respectively.

Based on the spatial distribution patterns of DIN , $\text{PO}_4\text{-P}$ and DSi , it could be concluded that the spatial distribution patterns of $\text{DIN}/\text{PO}_4\text{-P}$, DSi/DIN and $\text{DSi}/\text{PO}_4\text{-P}$ ratios in the inshore waters of the SWLZB were mainly associated with the distribution patterns of DIN and DSi concentrations.

In coastal waters, potential nutrient limitation could occur for P at $\text{Si}:\text{P} > 22$ and $\text{N}:\text{P} > 22$, for N at $\text{N}:\text{P} < 10$ and $\text{Si}:\text{N} > 1$, and for Si at $\text{Si}:\text{P} < 10$ and $\text{Si}:\text{N} < 1$ (Justić et al., 1995). Based on these criteria, the results of this study indicated that the inshore waters of the SWLZB, on the whole, were potentially P-limited, because the concentrations of dissolved nutrients in all the marine sampling sites showed a $\text{N}:\text{P}$ ratio of higher than 22:1 and they in more than half of the marine sampling sites showed a $\text{Si}:\text{P}$ ratio of higher than 22:1 (Fig. 6). Similar P limitation was also found in the inshore regions of the Changjiang Estuary (Chai et al., 2009), indicating that P limitation may be a common characteristic in river impacted coastal oceans of China. Nutrient acquisition is a major factor determining the outcome of competitive interactions and phytoplankton community composition (i.e., the species that make up the community). In P-limited ecosystems, ambient P concentrations governed phytoplankton growth rates and primary productivity (Goldman et al., 1979).

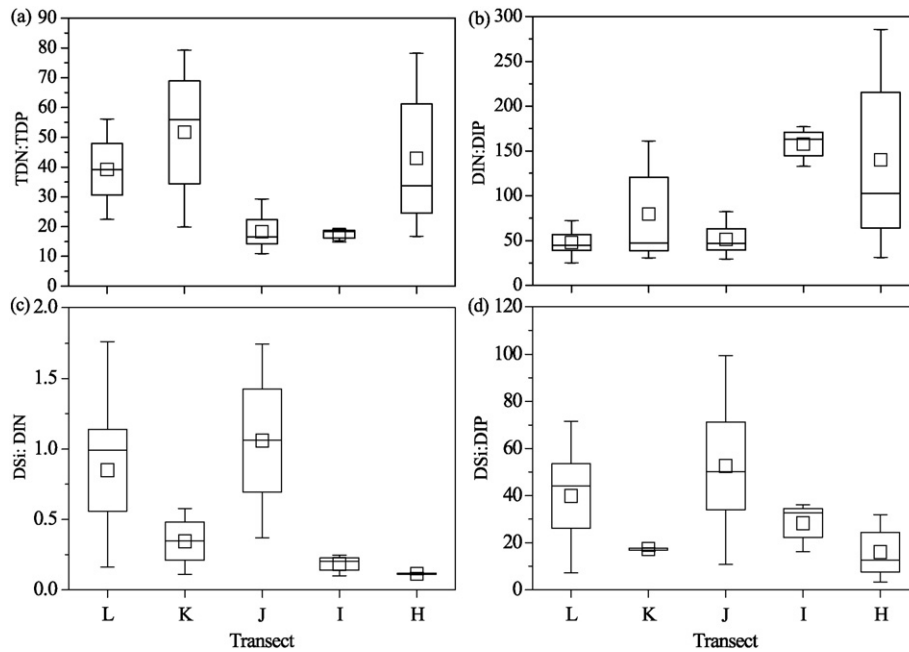


Fig. 5. Nutrient molar ratios in the inshore surface waters of the SWLZB.

3.4. Relationship between nutrients and acidification in the SWLZB

According to the above discussions, the N, P, Si concentration and structure in different sampling transects in the coastal waters of the SWLZB varied considerably (Fig. 4) depending on the specific contributors, and these variations were inevitably connected with the physico-chemical characteristics of the sampling waters. Here, Origin software was used to analyze the correlation between pH_T (as an acidification index) and the concentrations of TDN, DIN, DON, DSi, TDP, $\text{PO}_4\text{-P}$ and DOP, and the ratios of TDN/TDP, DIN/ $\text{PO}_4\text{-P}$, DSi/DIN and DSi/ $\text{PO}_4\text{-P}$. The results were shown in Figs. 7 and 8, which revealed that pH_T was significantly negatively correlated with the concentrations of $\text{NH}_4\text{-N}$, $\text{NO}_3\text{-N}$, DIN and DSi, in the surface waters of the SWLZB, but not correlated well with $\text{PO}_4\text{-P}$, DON and DOP concentration, and TDN/TDP, DSi/DIN and DIN/ $\text{PO}_4\text{-P}$ ratios, respectively. The general negative correlation between pH_T and nutrient concentration suggested the enrichment of some certain nutrients could enhance acidification in the inshore waters of the SWLZB. pH_T presented poor correlation with nutrient ratios, indicating that the changes in the structure of nutrients had limited influence on acidification.

3.4.1. Nutrient distribution and acidification

Correlations between pH_T and the concentrations of $\text{NH}_4\text{-N}$, $\text{NO}_3\text{-N}$, $\text{NO}_2\text{-N}$, DIN, DON and TDN were shown in Fig. 7a–f, as what can be seen that pH_T in the coastal waters of the SWLZB presented a significant

negative correlation with $\text{NH}_4\text{-N}$ ($P < 0.01$, $n = 48$), $\text{NO}_3\text{-N}$ ($P < 0.05$, $n = 48$), $\text{NO}_2\text{-N}$ ($P < 0.01$, $n = 48$), DIN ($P < 0.001$, $n = 48$), DON ($P < 0.05$, $n = 48$), and TDN ($P < 0.01$, $n = 48$), respectively. All the data in the coastal SWLZB revealed that higher dissolved nitrogen concentrations were accompanied with lower pH_T . Furthermore, because nitrogen speciation can also affect the cell-size distribution of phytoplankton communities, the larger species may have the capacity for more internal storage of nutrients and become dominant in fluctuating nutrient regimes (Margalef, 1978; Turpin and Harrison, 1979), and in general, smaller species have a higher preference for $\text{NH}_4\text{-N}$ uptake over $\text{NO}_3\text{-N}$ than larger phytoplankton species (Stolte et al., 1994), so different regions which have different sources and concentrations of N compounds may have different species in making up the phytoplankton community. Carbon dioxide (CO_2) released into the seawater in the respiration process of phytoplankton can lower the pH_T (that is, increasing acidity) of the subsurface waters, but different species of phytoplankton have different patterns and rates of respiration. The knock on effect was that the regions which have different sources and concentrations of N compounds may endure different degrees of acidification.

Correlations between pH_T and $\text{PO}_4\text{-P}$, DOP and TDP were shown in Fig. 7g–i. The data in the SWLZB revealed a negative effects of $\text{PO}_4\text{-P}$, DOP and TDP on pH_T , but all of them did not present close correlation with pH_T ($P > 0.05$, $n = 48$). Correlation between pH_T and DSi was shown in Fig. 7j. A significant negative correlation could be found between pH_T and DSi concentrations ($P < 0.001$, $n = 48$). The total nutrient concentrations (T nutrients = TDN + TDP + DSi) also showed a

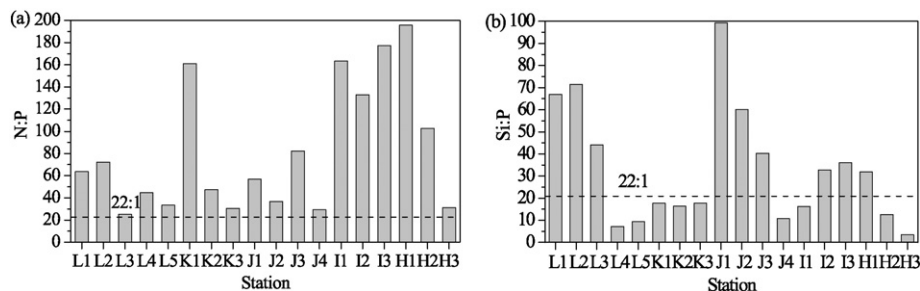


Fig. 6. Nutrient molar ratios in the inshore surface waters of the SWLZB. The dot lines indicate the nutrient molar ratios that were used to justice nutrient limitation for phytoplankton in coastal waters in Justić et al. (1995).

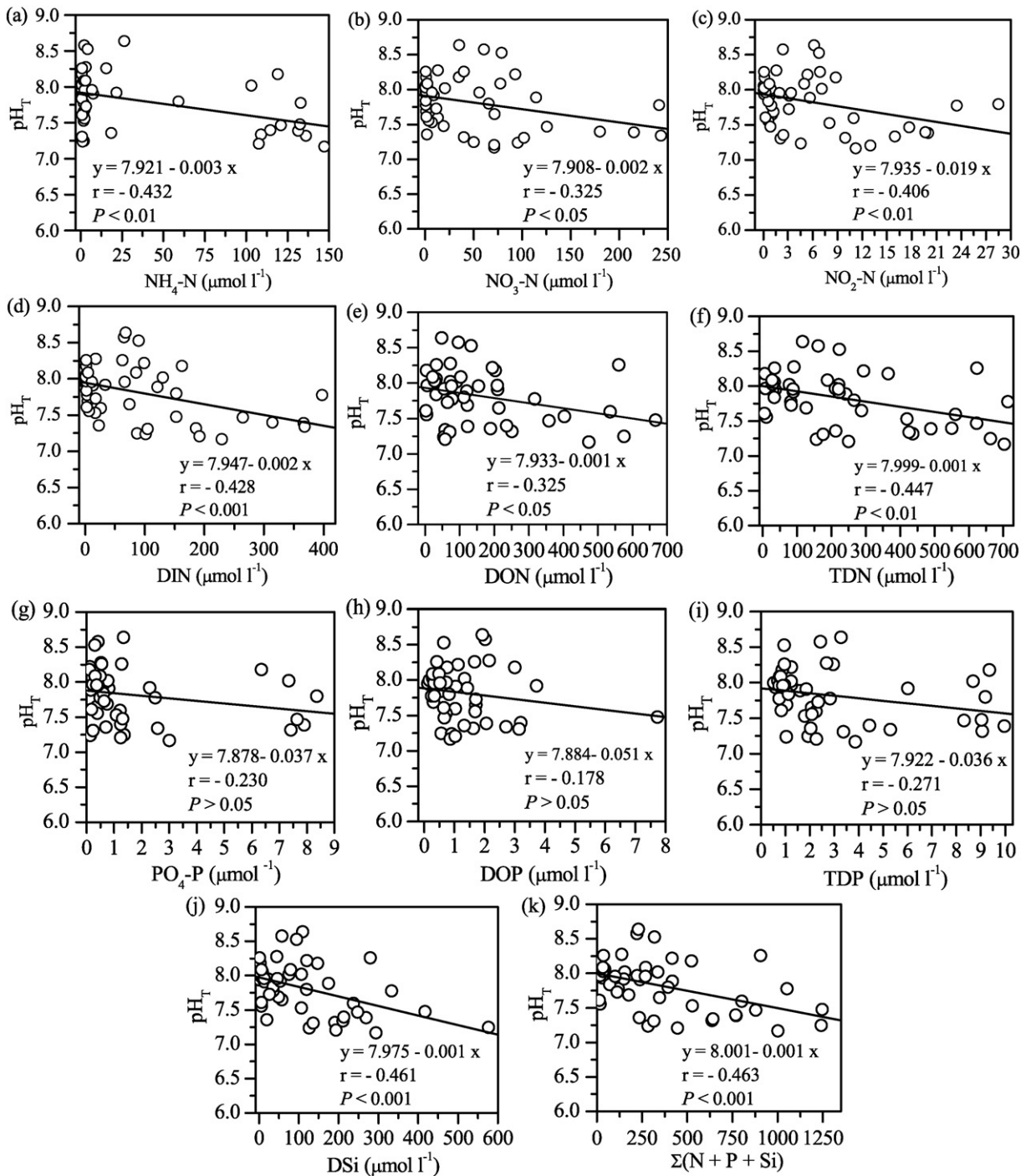


Fig. 7. Correlation between pH_T and nutrient concentrations.

significant negative correlation with pH_T ($P < 0.001$, $n = 48$) in the studied areas (Fig. 7k). Combining the above-mentioned data and discussions, it could be concluded that the enrichment of inorganic N and Si may enhance acidification in the SWLZB. Acidification is often accompanied with lower dissolved oxygen concentration, a common phenomenon of acidified regions related to eutrophication caused by nutrient enrichment (Chen et al., 2007; Gilbert et al., 2010), which in turn facilitates sediment nutrient release. Sediment nutrient release may further stimulate algal blooms, leading to a positive feedback loop that perpetuates blooms and eutrophication, subsequently to acidification.

However, the linkage between nutrient loading, eutrophication and hypoxia/anoxia dynamics is often non-linear and complex in estuarine and coastal systems (Cloern, 2001), so is to acidification. This is because

these systems are hydrodynamically and biogeochemically distinct and highly variable. Climatic and physiographic differences between these systems affect profoundly physical-chemical and biological processes mediating organic matter production and accumulation, oxygen dynamics, nutrient cycling and acidity.

3.4.2. Nutrient structure and acidification

Previous research has shown that phytoplankton species exhibit different growth responses to N sources (Collos, 1989). In particular, $\text{NO}_3\text{-N}$ and $\text{NH}_4\text{-N}$ uptake rates vary spatially and seasonally, suggesting differential community responses to the N sources (Boyer et al., 1994). These species-specific responses may play an important role in structuring natural phytoplankton communities (Stolte et al., 1994). Thus

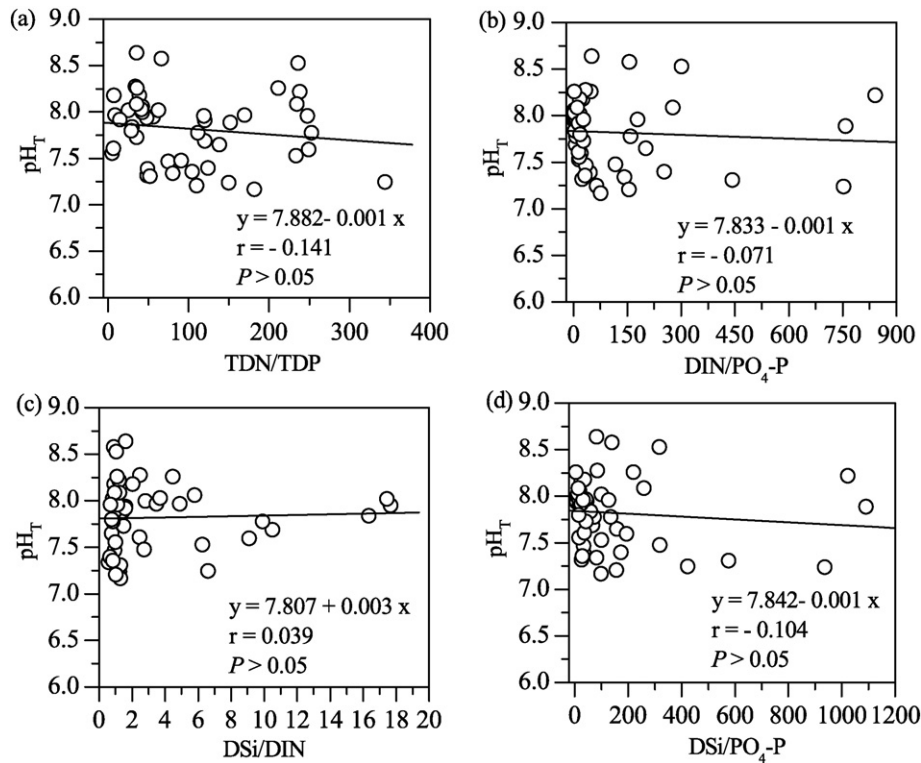
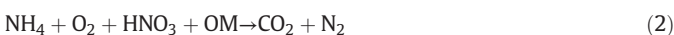


Fig. 8. Correlation between pH_T and nutrient molar ratios.

alteration of nutrient ratios may also change the phytoplankton community composition significantly, with subsequently knock on effect of alteration of acidification degree. Correlations between pH_T and TDN/TDP, DIN/ $\text{PO}_4\text{-P}$, DSi/DIN and DSi/ $\text{PO}_4\text{-P}$ ratios were shown in Fig. 8. In the present study, TDN/TDP, DSi/DIN, DSi/ $\text{PO}_4\text{-P}$ and DIN/ $\text{PO}_4\text{-P}$ ratios did not present close correlation with pH_T , respectively, suggesting that phytoplankton respiration was probably not the major reason for acidification in the SWLZB, and changing the nutrient ratios has limited effects on acidification in the SWLZB.

3.4.3. Mechanisms of nutrient affecting acidification

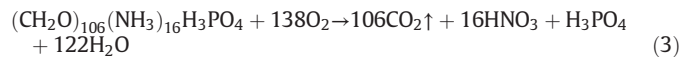
OA is directly related to CO_2 in the seawater; all the factors which can change the CO_2 concentration in the seawater could affect OA. Nutrient enrichment could enhance primary productivity which produces more organic matter (OM) in the seawater (Diaz and Rosenberg, 2008; Doney, 2010; Gilbert et al., 2010). The increase in primary production associated with nutrient enrichment will increase the concentrations of OM by two main mechanisms: secretion of mucopolysaccharides by phytoplankton, especially under conditions of stress, and bacterial degradation of phytoplankton-derived DOM and subsequent release of DOM and nutrients (Azam et al., 1983, 1993; Kirchman, 1991). DOM (e.g., glucose, urea, amino acids) may contain many inorganic nutrients required for the growth and metabolism of heterotrophic bacteria in aquatic systems (Azam et al., 1983, 1993; Kirchman et al., 1991). CO_2 released into the seawater in the respiration process of OM can lower the pH (that is, increasing acidity) of the subsurface waters in the same region. OM respiration can be illustrated as follows (Cai et al., 2011):



As can be seen in such a process not only CO_2 is produced but also O_2 consumed. So nutrient enrichment related eutrophication often causes low O_2 concentration. Biomass decomposition is often an O_2 -demanding process (Wallace et al., 2014). Therefore, waters enriched with

readily degraded or “reactive” organic matter tend to consume large amounts of O_2 . If the affected waters are vertically stratified, slowly flushed, and/or stagnant, consumption of O_2 may exceed its re-supply from either atmospheric or in-stream photosynthetic (i.e., O_2 evolution) sources. The imbalance between relatively high rates of O_2 consumption and low rates of O_2 re-supply causes DO concentration to drop to levels that are low enough to adversely affect oxygen-requiring animal and plant life (Diaz and Rosenberg, 1995; Wallace et al., 2014).

Respiration of OM below the pycnocline introduces additional CO_2 into the subsurface waters, hence reducing the pH and seawater carbonate saturation state. When O_2 is available, the most important respiration process is as follows (Cai et al., 2011):



Note that on hydration CO_2 becomes carbonic acid and the production of a small amount of the strong acid, HNO_3 , may play a role to some extent in reducing pH and total alkalinity (TA) (Canfield et al., 1993; Lohrenz et al., 2010).

In the aquatic system, $\text{NO}_3\text{-N}$ and $\text{NH}_4\text{-N}$ may perform as weak acid and base, respectively, and different phytoplankton communities consume different amounts of $\text{NO}_3\text{-N}$ and $\text{NH}_4\text{-N}$ (Stolte et al., 1994), resulting in $\text{NO}_3\text{-N}$ presenting negative correlation with pH_T (Fig. 7d, enhance acidification) and $\text{NH}_4\text{-N}$ presenting negative correlation with pH_T (Fig. 7f, mitigate acidification). Some composition of silicate minerals in water system may act as weak acid (e.g. Na_2SiO_3 , K_2SiO_3 , KNaSiO_3 , etc.); furthermore, DSi supports the major phytoplankton community—diatom in coastal water system, (Stolte et al., 1994; Diaz and Rosenberg, 2008; Doney, 2010; Gilbert et al., 2010), which release CO_2 into seawater during respiration. Therefore, higher DSi concentration was accompanied with lower pH_T value in this research (Fig. 7j, c, d, enhance acidification).

Since P was the potential nutrient limitation of phytoplankton growth in the SWLZB, primary productivity in this region may be sensitive to $\text{PO}_4\text{-P}$ concentrations, and higher $\text{PO}_4\text{-P}$ concentrations could

induce higher OM concentrations, consequently inducing lower pH values. However, the pH_T presented poor correlation with $\text{PO}_4\text{-P}$ concentrations (Fig. 7g), suggesting that the input of acid materials (e.g. $\text{NO}_3\text{-N}$, Na_2SiO_3 , K_2SiO_3 , etc.) rather than OM respiration may be the major reasons for acidification in the SWLZB.

4. Conclusion

In this research, the pH_T values in the inshore waters of the SWLZB were found to vary from 7.56 to 8.26 with an average of 7.93, which was significantly below the mean pH_T value of the world oceans, and the decadal change of pH values in the bay demonstrates a decreasing trend, indicating a sign of acidification, suggesting that the SWLZB are experiencing acidification.

The dissolved nitrogen in the SWLZB varied greatly along the coast corresponding to different contributors. It decreased significantly in the seaward direction. The SWLZB was enriched with nitrogen. The distribution pattern of nutrients in the marine sampling transects is generally similar to that of nutrients in their corresponding rivers which indicated that the transportation and behavior of nutrients in the SWLZB were heavily influenced by river inputs and anthropogenic activities. The general DIN structure in the L-transect and the K-transect was $\text{NH}_4\text{-N} > \text{NO}_3\text{-N} > \text{NO}_2\text{-N}$; in the J-transect, the I-transect and the H-transect, it was $\text{NO}_3\text{-N} > \text{NH}_4\text{-N} > \text{NO}_2\text{-N}$; the average concentrations of $\text{NO}_2\text{-N}$ in the five sampling transects remained at the same level, indicating that $\text{NO}_2\text{-N}$ in the studied region was very likely to be from non-point sources.

The concentration of dissolved phosphorus decreased from the estuaries to the sea; the average concentrations of reactive phosphorus in the five marine sampling transects were relatively low. The inshore waters of the SWLZB had high nitrogen concentrations and low phosphorus concentrations. There were >70% of the sampling sites exhibiting a condition of potential P limitation in the inshore waters of the SWLZB.

In the SWLZB, pH_T was significantly negatively correlated with the concentration of DIN, $\text{NO}_3\text{-N}$, DSI and total nutrient, respectively. The negative correlation between pH_T and nutrients suggested the enrichment of some certain nutrients enhanced acidification in the SWLZB. In the P limitation circumstances more inorganic P would be beneficial for some kinds of biomass, and more CO_2 will be released into seawater. But pH_T in the SWLZB presented poor correlation with $\text{PO}_4\text{-P}$ concentration and nutrient ratios, suggesting that the input of acid materials (e.g. $\text{NO}_3\text{-N}$, Na_2SiO_3 , K_2SiO_3 , etc.) rather than OM respiration may be the major reasons for acidification in the inshore waters of the SWLZB.

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